

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re the Application of: Makoto IWAI, Katsuhiko IMAI and Minoru IMAEDA

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Confirmation No.: 3664

FOR: GALLIUM NITRIDE SINGLE CRYSTAL GROWING METHOD AND
GALLIUM NITRIDE SINGLE CRYSTAL

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

RULE 132 DECLARATION OF MAKOTO IWAI

Sir:

I, Makoto Iwai, a citizen of Japan, hereby declare and state that:

1. I have a Ph.D degree in electrical engineering which was conferred upon me by Osaka University in Japan, in 1997.
2. I have been employed by NGK Insulators, Ltd., the Assignee of the above-identified application, since 1997, during which time I have been involved in the research and development of materials for opto-electronics.
3. I am a member of The Japan Society of Applied Physics and The Japanese Association for Crystal Growth.
4. I am one of the inventors of the above-referenced application, and am familiar with the prosecution history, including the Office Action mailed February 25,

2010, and the prior art references applied by the Examiner therein, namely, Sasaki, Sarayama, and D'Evelyn.

5. Our invention relates to a method of growing a gallium nitride single crystal using the sodium metal flux technique. The conventional wisdom for that technique states that it should be carried out at a relatively low nitrogen pressure (e.g., around 50 atms) and at a relatively low temperature (e.g., around 750 - 800°C). Please see, for example, paragraph [0158] of the Sarayama reference applied by the Examiner.

6. Mr. Sarayama, who is employed by Ricoh Co., Ltd., is a well-known researcher in sodium metal flux techniques, as evidenced by his publication attached hereto as Appendix A, the abstract of which again confirms that the sodium metal flux technique is performed at a relatively low nitrogen pressure (e.g., about 50 atms (5 MPa)) and at a relatively low temperature (e.g., 800°C).

7. Attached hereto as Appendix B is a copy of U.S. Patent No. 7,524,691, which is owned by another company, Panasonic Corporation, and also confirms that the sodium metal flux technique is carried out at a relatively low pressure (50 atms) and at a relatively low temperature (750°C), similar to those disclosed in the Sarayama reference applied by the Examiner (see U.S. '691 at Col. 1., lines 39-49).

8. I can personally confirm what these documents plainly state -- the conventional wisdom associated with the sodium metal flux technique has been that it should be carried out at relatively low pressures (e.g., around 50 atms) and at relatively low temperatures (e.g., 750 - 800°C).

9. Our claimed invention defies the conventional wisdom. We carry out the sodium metal flux technique at a substantially higher pressure of 300 atms to 1200 atms and at a substantially higher temperature of 850°C to 1200°C. As a result, we have been able to achieve gallium nitride single crystal growth rates of at least 25 $\mu\text{m/hr}$ (please see Example 2 in the present specification), which is a three-fold improvement over the 8.3 $\mu\text{m/hr}$ growth rate achieved by the method disclosed in Sasaki (see Example 7, paragraph [0077] of Sasaki). We have even been able to achieve growth rates of 50 $\mu\text{m/hr}$, which is a six-fold improvement over the process disclosed in Sasaki (please see Examples 1 and 5 in the present specification).

10. The presently claimed invention was not obvious to me at the time it was made, because again, our approach defies the conventional wisdom associated with the sodium metal flux technique used for growing gallium nitride single crystals.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

June 17, 2010
Date

Makoto Iwai
Makoto Iwai

Appendix A - Sarayama et al., *High Quality Crystal Growth of Gallium Nitride by Flux Method*, Ricoh Technical Report No. 30, December 2004

Appendix B - U.S. Patent No. 7,524,691

フラックス法による高品質窒化ガリウムの結晶成長

High Quality Crystal Growth of Gallium Nitride by Flux Method

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岩田 浩和*

Hirokazu IWATA

要 旨

ナトリウム (Na) フラックスを用いた窒化ガリウム (GaN) 結晶成長技術の研究開発を進めている。本フラックス法では温度800℃、窒素圧力5MPa程度の成長条件下で高品質GaN結晶が成長可能となる。成長温度と窒素圧力を選ぶことにより、GaN結晶の分解と種結晶成長及び自発核成長の各成長モードを制御し、更に自発核成長領域では板状結晶と柱状結晶の形態制御を実現した。

自発核成長モードにおいて、板状結晶で転位密度 $<10^6\text{cm}^{-2}$ 、柱状結晶では観察出来ない程度の低転位密度の超高品質結晶を実現できた。結晶サイズは板状結晶で5~10mm程度、柱状結晶で長さ5mm・径1mm程度である。種結晶成長モードでは、リチウム (Li) 添加により表面平坦性を数 $10\mu\text{m}$ の凹凸から $1\mu\text{m}$ 以下の凹凸にまで向上させることが出来た。これらの結果は、他方式である気相成長法に比較して品質面で優れており、安熱法や高圧溶液法等の他の液相成長方式に比較して結晶サイズは大きい。

ABSTRACT

We have studied the growth technique of Gallium Nitride (GaN) single crystals using Sodium (Na) (Flux Method). In this Flux Method, high quality GaN crystals can be grown under the condition of proper temperature (about 800℃) and N_2 pressure (about 5MPa). Growth modes having decomposition, seed growth and self-nucleation growth can be controlled by the change of temperature and N_2 pressure, and moreover hexagonal platelet and prismatic crystals are obtained.

Dislocation density of the platelet crystals is less than 10^6cm^{-2} and that of the prismatic crystals is below measurable limit in the self-nucleation growth mode. The size of the platelet crystals is 5~10mm and the length and the diameter of the prismatic crystals are 5mm and 1mm respectively. By adding Lithium (Li) the surface roughness of GaN crystals can be improved to less than $1\mu\text{m}$ from tens of micrometers. These GaN crystals have higher quality than crystals grown by Vapor Phase Epitaxy Method, and are larger than crystals grown by other liquid phase growth methods (e.g. Ammonothermal Method and Super High Pressure Solution Method, etc.)

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1. 背景と目的

1-1 背景

近年、青色発光ダイオード（LED）や白色LEDや青紫色半導体レーザ（LD）など、窒化ガリウム（GaN）系材料を用いた半導体デバイスの応用進展が著しい。応用機器として、屋外用途では交通信号機や大画面ディスプレイ、パーソナル機器として携帯電話画面のバックライトやブルーレイディスク等の高密度光メモリ用光源等が挙げられる。

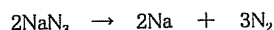
これらの現在実用化しているGaN系半導体デバイスは、主にサファイアやシリコンカーバイド（SiC）を基板として、その上にGaN系薄膜を有機金属気相成長（MOVPE）法や分子線エピタキシー（MBE）法によりヘテロエピタキシャル成長することにより実現している。この場合、基板とGaN系薄膜の熱膨張係数と格子定数が大きく異なるため、 10^9cm^{-2} 以上の高密度の転位がGaN系薄膜に生じる¹⁾。A. Usuiらは、サファイア基板上にGaN結晶を気相成長により横方向選択成長（ELO）させ、欠陥密度を低減させた²⁾。その後、このELO技術を用いてGaAs基板³⁾やサファイア基板⁴⁾上に数 $100\mu\text{m}$ のGaN厚膜結晶を成長させた後、基板剥離することにより、転位密度 $10^6\sim 10^7\text{cm}^{-2}$ 程度のGaN厚膜基板が得られているが、依然としてヘテロエピタキシャル成長の限界を大きく超える低転位密度化は実現出来ていない。この方式は一般にVPE（Vapor Phase Epitaxy）方式と呼ばれている（Table 1参照）。

一方、ポスト蛍光灯と目されている高演色性の白色LEDや高速高密度光メモリ用青紫色LD実現のためには、転位密度 10^4cm^{-2} 以下の高品質GaN基板の開発が必要である^{5,6)}。高品質GaN基板の実現を目指して、Table 1に示す各種技術が様々な研究機関で開発されている。

Table 1に示すように、GaNバルク結晶成長の手法は気相成長と液相成長に大別される。気相成長には、前述したVPE方式、Ga（ガリウム）蒸気とアンモニアや窒素ガスから直接合成する気相合成、GaN原料を昇華させ成長する昇華法がある。液相成長には、6万気圧で 2200°C の条件下でGaNを融液化して再結晶化させる融液成長と、溶媒に窒素やGaNを溶解させ成長させる溶液成長がある。溶液成長には、高温（ 1600°C ）のGa融液中に超高圧（ $1\sim 2$ 万気圧）の窒素ガス下で窒素を溶解させGaNを成長する高圧溶液法、本報告のフ

ラックス法、液体若しくは超臨界アンモニア中（数千気圧、数 100°C ）にGaNを溶解させ再結晶化させる安熱法がある。この内、VPE法は基板として $\phi 2$ インチ以上のサファイアやGaAsを用いることで、実用サイズの結晶が得られているが、基本的にはヘテロエピタキシャル成長であることから、前述したように 10^6cm^{-2} 程度の欠陥が残る。気相合成や昇華法ではサブmmオーダーの結晶しか実現していない。液相成長は気相成長に比較して高品質な結晶が得られるが、一般的に結晶サイズは小さい。融液成長と安熱法では結晶サイズは $100\mu\text{m}\sim 1\text{mm}$ である。高圧溶液法ではmmオーダーの結晶が得られているが、その厚さは数 $10\mu\text{m}$ と薄く、取り扱いが困難である。更に融液成長、高圧溶液法、安熱法では圧力あるいは温度が高いために、量産性に課題がある。

一方、フラックス法は1997年にH. Yamaneらにより初めて報告された結晶成長方法であり¹³⁾、アジ化ナトリウム（ NaN_3 ）と金属Gaを $600\sim 800^\circ\text{C}$ の温度に加熱することで、結晶サイズ $500\mu\text{m}$ 程度のGaN結晶が成長する。このとき 300°C 以上の温度で、



となり、成長温度では窒素圧力 $<10\text{MPa}$ と推定され、他の液相成長に比較して低温低圧での結晶成長が可能である。

Table 1 Bulk GaN crystal growth methods

方式		特徴
気相	VPE (Vapor Phase Epitaxy)	サファイアやGaAs等の基板上に気相により選択成長，厚膜化して転位密度 10^8cm^{-2} 台，結晶サイズは $\phi 2$ インチを実現 ^{1, 4)} 。
	気相合成 ⁷⁾	1200℃以上の高温下で金属Gaの蒸気とNH ₃ ，N ₂ からGaNを成長させる。結晶サイズ<1mm
	昇華法 ⁸⁾	GaN粉末を1500℃前後で昇華させ，低温部の種結晶に結晶成長させる。結晶サイズ<1mm
液相	融液成長 ⁹⁾	6GPa，2200℃以上の高温高窒素圧力化で徐冷し，100 μm 程度のGaN結晶が成長。
	高圧溶液法 ^{10, 11)}	1~2GPa，1600℃の高温高圧下で金属Ga中に窒素を溶解させ，mmオーダーの結晶が成長。
	フラックス法	NaやK等のアルカリ金属とGaからなる混合融液に窒素圧力<10MPa，温度800℃前後で数mmのGaN結晶が成長。
	安熱法 ¹²⁾	超臨界若しくは亜臨界のNH ₃ （）にGaN原料を溶解，再結晶化させる。結晶サイズは100 μm ~1mm。

1-2 目的

以上述べたように、現時点ではデバイスニーズに適応可能な高品質（転位密度 10^4cm^{-2} 以下）でインチオーダーのGaNバルク結晶は存在しない。筆者らは実用的な結晶成長条件で高品質な単結晶が得られるGaN結晶の大型化を目指して研究

開発を行っている。本研究では、前述したH. Yamaneらによって開発されたフラックス法を更に発展させ、窒素源として N_2 （窒素ガス）及びフラックスとして金属Na（ナトリウム）を用いて、大型化と高品質化の両立を図っている。

本報告では、東北大学山根教授らとの共同研究結果を引用しながら、これまで得られた研究結果について述べる。

2. 技術

2-1 フラックス法とは

2-1-1 成長メカニズム

フラックス法によるGaN結晶成長は、高压溶液法との対比で次のように説明出来る。

Fig.1は、Karpinskiらが報告したGaNの平衡窒素圧力と温度の関係¹⁴⁾に、フラックス法によるGaN結晶成長条件を追記したものである¹⁵⁾。高压溶液法及びフラックス法の何れもGaN安定領域で成長していることが判る。

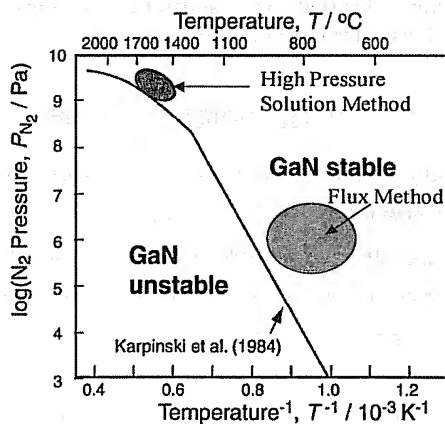


Fig.1 Solution growth phase diagram of GaN crystal.

高压溶液法では金属Gaに窒素を溶解させるために、1500～1600℃の温度と1GPa以上の N_2 圧力を必要としている。フラックス法ではNaを用いることで、GaN結晶成長に必要な温度、圧力を各々800℃前後、<100MPaに低減することができる。Naの役割はFig.2で示すように考えられている¹⁵⁾。

アルカリ金属であるNaは電気陰性度やイオン化エネル

ギーが小さいために、NaとGaの混合融液表面の N_2 分子へ電子を移動し易い。 N_2 分子は結合性分子軌道まで電子が埋まっていることから、Naから移動した電子は反結合性軌道に入り、三重結合である N_2 の結合力は低下する。この結果、Na-Ga混合融液中にNが溶解することとなる。溶解したNは、 $(Ga_xN_y) \cdot (Na_z)$ の可溶種を形成し、GaNの過飽和状態の領域（坩堝内壁やGaN結晶表面）においてGaN結晶として成長することとなる。

このようにNaがあることで、超高压と高温の成長条件を低圧低温にすることが可能となることから、フラックス法と呼ばれている。またフラックスとしてはNaのみならず、他のアルカリ金属としてK（カリウム）を用いることでもGaN結晶が成長可能であることも明らかとなっている¹⁶⁾。

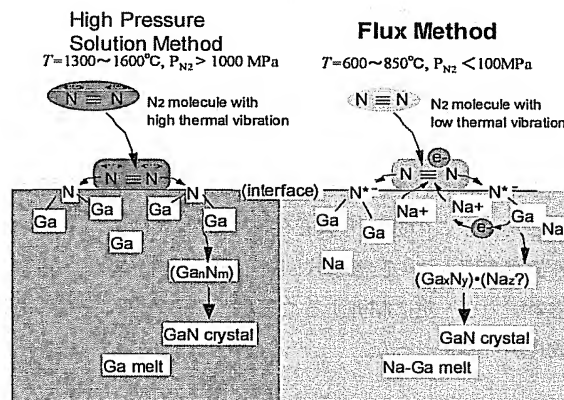


Fig.2 GaN crystal growth model by Na Flux Method.

2-1-2 結晶成長装置～原料供給と加熱方式の検討

当初のフラックス法は、前述したように NaN_3 と金属Gaをステンレスチューブの反応容器に封入し、電気炉内で加熱することでGaN結晶を成長していた。しかしこの場合には、 NaN_3 の分解により生じる窒素が原料となるため、窒素の圧力変動や枯渇により、結晶サイズ $\leq 1\text{mm}$ 程度にとどまっていた。更に反応容器からのCr等の溶出、原料中の不純物の問題もあった。

そこで我々は山根らと共同で、原料を高純度窒素ガス（6N up^{*}）と高純度金属Ga（7N）及びフラックスとして高

* 6N up: 純度>99.9999%, 7N: 純度>99.99999%

純度金属Na (3N up : >99.95%) を用い、BN (窒化ホウ素 : ボロンナイトライド) を坩堝として用いた結晶成長方法を新たに開発した。このとき窒素ガスの導入方法¹⁷⁾と坩堝形状の工夫¹⁸⁾、及びNa純度の結晶成長に及ぼす影響¹⁹⁾を明らかにすることにより結晶サイズ拡大を可能とした。これらの経緯¹⁵⁾をFig.3に示す。

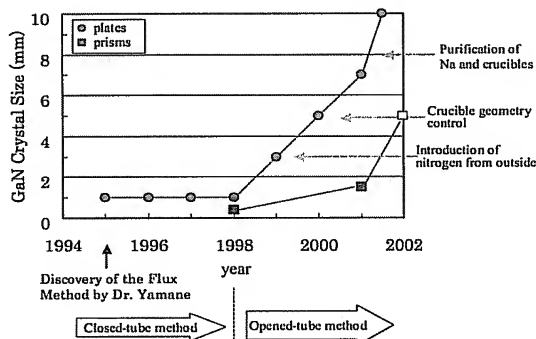


Fig.3 Progress of GaN crystal size.

このように Na_3N を用いずに外部から窒素ガスを導入することから、この成長方式を外部窒素導入法と呼ぶ。Fig.4に外部窒素導入法の装置模式図を示す。外部窒素導入法により窒素圧力を一定に制御することが可能となり、安定した結晶成長が可能となった。

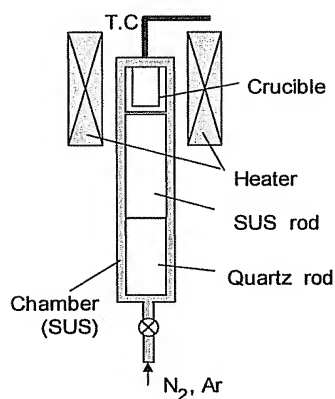


Fig.4 Outer heating type apparatus for Flux Method. Nitrogen is introduced from outside.

これまで説明した結晶成長で用いた装置は、外径1/2~1インチのステンレスチューブを反応容器とし、この反応容器を電気炉で加熱する構成となっていた。この加熱方式を外熱

方式と呼ぶ。外熱方式では耐圧性と耐熱性の点で、内部に入れる坩堝サイズに制限(外径20mm程度)があった。結晶の更なる大型化のためには坩堝サイズを大きくする必要があり、新たに内熱方式と呼ばれる加熱方式を開発した(Fig.5)。内熱方式では耐圧性と耐熱性をそれぞれアウターチャンバーとインナーチャンバーに機能分離することにより、坩堝サイズを外径40mm程度に拡大することが可能となった。この結果、2-2で後述するように柱状のGaN結晶のサイズ拡大を実現した。

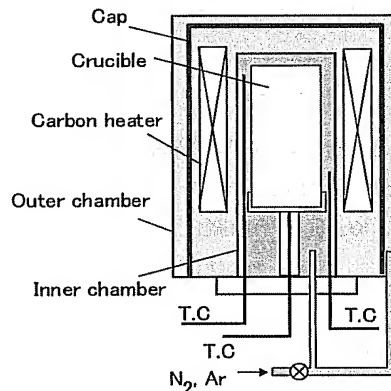


Fig.5 Inner heating type apparatus for Flux Method. Nitrogen is introduced from outside.

2-2 フラックス法での成長モード制御

外部窒素導入法により窒素圧力と成長温度を独立制御することが可能となり、フラックス法における窒素圧力と成長温度がGaN結晶に与える影響を調べた²⁰⁾。Fig.6にその結果を示す。窒素圧力が低く成長温度が高い領域(領域③)では、GaN結晶は成長せずに溶解する。一方、窒素圧力が高く成長温度が低い領域(領域①)では、多核成長モードとなる。この二つの領域の間の領域(領域②)で、種結晶にのみ成長する種結晶成長モードがあることが判った。

このことはNa-Ga混合融液中のGaNの過飽和度に依存して、成長モードが変わると考えると説明できる。初期の状態から温度と窒素圧力が所定の状態になると、混合融液中への窒素の溶解が始まる。このとき混合融液中の温度が低く、窒素圧力が高い場合には、混合融液のGaN飽和溶解量は小さいが窒素溶解速度は大きく局部的に過飽和度が大きくなり多核発生し易くなると考えられる。この状態が領域①であり、多数の

結晶核が発生し、その結晶核に対してGaN結晶成長が進行して、各結晶が大きくなっていくと考えられる。領域②ではGaNの過飽和度が小さくなり、自発核が発生せずに種結晶領域にのみGaN結晶が成長すると考えられる。更に領域③では、混合融液中への窒素の溶解度が小さくGaN結晶が成長せず、GaN結晶が混合融液中に存在する場合には、このGaN結晶が溶解するものと考えられる。

また、領域①の多核成長モード内で、窒素圧力が高く成長温度が低い領域（領域①-1）では板状結晶が支配的に成長し、窒素圧力が低く成長温度が高い領域（領域①-2）では柱状結晶が支配的に成長することも判っている。板状結晶の主面はC面であり、柱状結晶の軸方向はC軸であることが判っていることから、窒素の溶解度が高いほどC面が支配的な成長が促進されるものと考えられる。

領域③においてGaNが溶解することから、水晶の結晶成長で用いられる水熱合成と同様に、溶解再結晶化によるGaN結晶成長も可能である²¹⁾。

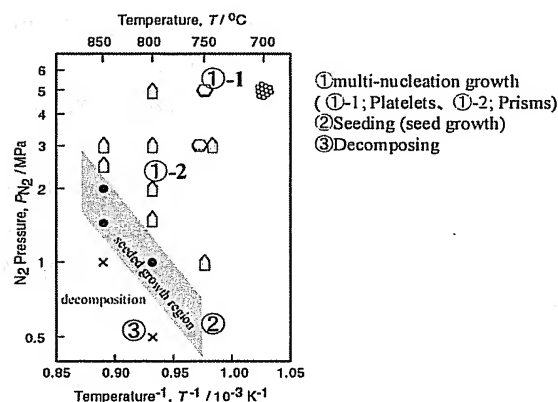


Fig.6 Flux Method P_{N_2} -T diagram of GaN crystal.

以上のように、窒素圧力と温度を制御することにより、自発核発生による多核成長、種結晶成長及び溶解再結晶化の成長モード制御が可能であり、更には多核成長において板状結晶と柱状結晶の結晶形態制御が可能となった。以下の2-3および2-4において、各領域で得られるGaN結晶について詳説する。尚、多核成長モードでの結晶成長を自発核成長と呼ぶ。

2-3 自発核成長

2-3-1 実験方法

Fig.4あるいはFig.5の結晶成長装置内に設置した坩堝の断面模式図をFig.7に示す。Fig.7(a)は結晶成長開始前の状態を示しており、BN坩堝内にNa-Ga混合融液が保持されている。この状態で温度700～800℃、 N_2 圧力3～8MPa、成長時間100～300時間保持することで、Fig.7(b)のように坩堝内にGaNが自発核成長する。

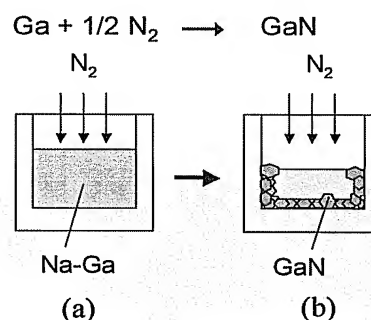


Fig.7 Cross-sectional view of the crucible shown in Fig.4 or Fig.5.

2-3-2 結晶形態

Fig.8に自発核成長した結晶の顕微鏡写真を示す。(a)は板状結晶であり、(b)は柱状結晶である。それぞれ窒素圧力(a) 4MPa、(b) 8MPa、成長温度(a) 775℃、(b) 800℃であり、Fig.6の①-1と①-2の領域に対応している。

(a)の板状結晶は径3～4mmで、その主面はC面即ち(0001)面であり、六角板形状となっている。一方(b)の柱状結晶は径0.5mmで長さ2mmであり底面がC面の六角柱形状となっており、先端が結晶面方位{10-11}の斜めのファセットで形成されている。

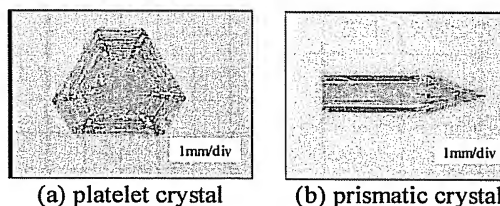


Fig.8 Micrographs of GaN crystals grown by Flux Method at the self nucleate growth mode.

以上、結晶形態の成長温度と窒素圧力依存性について述べたが、次にフラックス法による結晶成長のもう一つのキーパラメータであるNaモル量比の結晶形態に与える影響について述べる。

成長温度775℃、窒素圧力4MPa、成長時間300時間と一定にし、Naモル量比 r_{Na} を0.4, 0.62, 0.67と変化させた。ここで、Naモル量比 r_{Na} は次の式で表される。

$$r_{Na} = [Na] / ([Na] + [Ga])$$

ここで[Na]と[Ga]はそれぞれNaとGaのモル量である。

得られた結晶の顕微鏡写真をFig.9に示す。Naモル量比が大きくなることで板状結晶が成長し易い、即ちC面が支配的な成長となる。一方、Naモル量比が小さくなると柱状結晶や先端が結晶面方位 $\{10\bar{1}1\}$ の斜めのファセットが成長し易くなる。

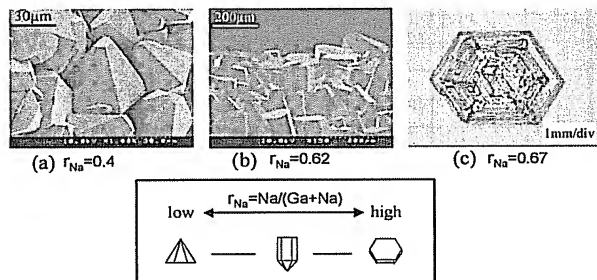


Fig.9 Micrographs showing morphological changes of GaN crystals at various r_{Na} s.

以上の結晶形態に関してまとめると、窒素圧力及びNaモル量比が大きい場合C面が支配的な成長となり、窒素圧力及びNaモル量比が小さくなるに従い、c軸方向の成長速度が速くなり、c軸に平行な $\{10\bar{1}0\}$ 面やc軸に対して30°傾いた $\{10\bar{1}1\}$ 面が現れる。この模式図をFig.10に示す。

この結晶形態の変化には2-2で既に述べたように、Na-Ga混合融液中の窒素の溶解度が影響していると考えられる。窒素圧力やNaモル量比が大きいほど、混合融液中への窒素の溶解度が高くなる。この結果、c面方向の成長速度が早くなるものと考えられる。

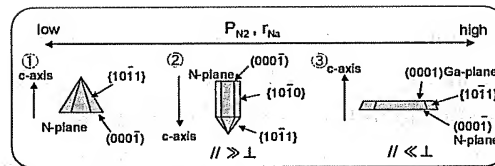


Fig.10 Schematic model for morphology change of GaN crystals at various P_{N_2} and r_{Na} .

2-3-3 坩堝サイズ効果

この自発核成長において結晶サイズを拡大するために、幾何学的選別作用²²⁾の適用を検討した。幾何学的選別作用は鉱物や単結晶成長等で多核発生し、その後の継続的な結晶成長において優先核が生き残り、結晶サイズが拡大するものである。そのためには結晶成長が継続する必要がある。本フラックス法の場合には、Na-Gaの混合融液量を増やすこと及び結晶が大きく成長しても十分な空間が確保されている必要がある。そこで2-1-2で述べた内熱方式の結晶成長装置を用いて、坩堝サイズを拡大し、柱状結晶の成長モードにて結晶成長を行った。

その結果をFig.11に示す。坩堝A, B, Cはそれぞれ内径φ16, 30, 37mmである。それに対して成長した結晶サイズは、坩堝Aでは径0.1mm, 長さ0.5~1mm, 坩堝Bでは径0.3mm, 長さ1~2mm, 坩堝Cでは径0.7mm, 長さ3~5mmとなっている。明らかに坩堝サイズ拡大に伴い、結晶サイズも拡大していることが判る。この結果から、フラックス法によるGaN結晶成長の自発核成長モードにおいても、幾何学的選別作用が働き、優先核が成長することが考えられる。

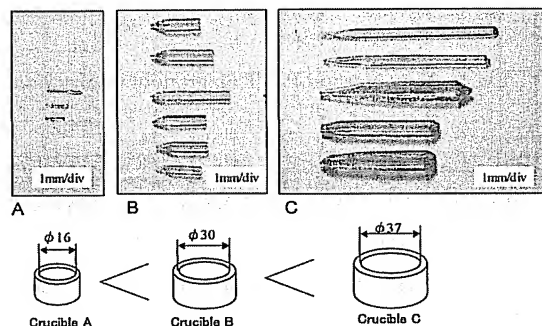


Fig.11 Photographs of obtained prismatic crystals at different crucible sizes.

2-3-4 結晶性評価

自発核成長で得られた板状結晶のエッチピット密度 (EPD: Etch Pits Density) 評価を行った。エッチング液は硫酸 (H_2SO_4) と磷酸 (H_3PO_4) の混酸 (1:1), 230°C である。エッチング後のエッチピットの顕微鏡写真をFig.12²³⁾に示す。リファレンスとして、サファイア基板上にMOVPE (Metal Organic Vapor Phase Epitaxy) 法によりGa₂N薄膜を約1 μm 成長したものを用い、これら二つの試料は同時にエッチングした。

Fig.12 (a)はフラックス法により成長したGa₂N板状結晶で、(b)がサファイア基板上のGa₂N薄膜である。試料(a)のGa₂N板状結晶はステップとテラス領域があり、ステップ近傍にエッチピットが観察されるが、テラスには殆どエッチピットは観察されない。このエッチピットが観察されるステップ近傍でもEPDは 10^5cm^{-2} 台である。一方、試料(b)のGa₂N薄膜のEPDは 10^8cm^{-2} 台と高密度の結晶欠陥を有していることが判る。

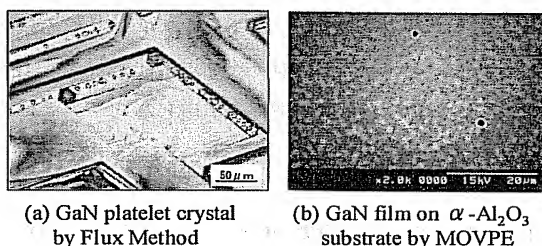


Fig.12 Photographs of etch pits in GaN crystal.

以上のことからフラックス法により成長した板状結晶は、ステップ近傍の結晶欠陥の多い領域においても、サファイア基板上に成長したGa₂N薄膜と比較して、数桁小さいEPDとなっており、高品質であると言える。

次に板状結晶のTEM (透過型電子線顕微鏡) 観察結果をFig.13に示す。Fig.13 (a)を拡大したものが(b)である。この板状結晶は平坦化するために、表面を研磨している。そのため表面から1 μm の領域で研磨によるダメージ層が生じており、それが転位となって観察されている。一方、このダメージ層より深い領域では転位は観察されていない。このことから本来フラックス法で自発核成長したGa₂N板状結晶の転位密度は、 $<10^6\text{cm}^{-2}$ と見積られる。この結果は、EPDの結果とほぼ一致している。

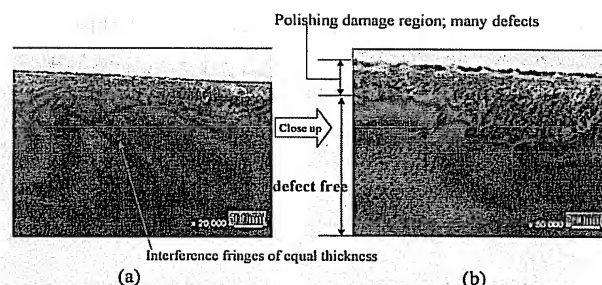


Fig.13 TEM images of platelet crystal after polishing.

次に板状結晶の特性評価結果を高圧溶液法^{10, 11)}と比較してTable 2に示す²⁴⁾。板状結晶のX線ロッキングカーブの半値全幅は45~55arcsecである。電気的特性は導電型がn型であり、電気抵抗率は0.04 Ωcm 、キャリア濃度 $n = 1 \sim 2 \times 10^{18}\text{cm}^{-3}$ 、移動度 $100\text{cm}^2\text{V}^{-1}\text{s}^{-1}$ である。何れの値もこれまで最も高品質なバルク結晶と言われてきた高圧溶液法と比較して、結晶性の点で同等あるいはそれ以上であると言える。

Table 2 Properties of GaN platelet single crystals

	the present study	High Pressure Solution Method
FWHM of X-ray rocking curve (0002)	45 ~ 55 arcsec (6 ~ 10 mm size)	30 ~ 120 arcsec.
electrical properties		
(RT) electrical resistivity	0.04 Ωcm	—
carrier concentration (n-type)	$1 \sim 2 \times 10^{18}\text{cm}^{-3}$	$10^{19} \sim 10^{20}\text{cm}^{-3}$
mobility	$100\text{cm}^2\text{V}^{-1}\text{s}^{-1}$	$30 \sim 90\text{cm}^2\text{V}^{-1}\text{s}^{-1}$

次にフラックス法により成長した板状結晶と柱状結晶のカソードルミネッセンス観察結果をFig.14に示す。この写真は電子線をエリヤスキャンして、波長365nmの発光を分光して観察している。転位は非発光の再結合センターとなることから、転位が存在する領域ではダークスポットとして観察される。従ってダークスポットから転位密度を評価することが出来る。

Fig.14 (a)は板状結晶、(b)は柱状結晶のそれぞれのカソードルミネッセンス像である。(a)の板状結晶ではダークスポットが数個観察され、ダークスポット密度は約 $2 \times 10^5\text{cm}^{-2}$ である。この値は前述したEPD評価の値とほぼ一致する。一方(b)の柱状結晶において、ダークスポットは観察されないことから、板状結晶と比較しても高品質な結晶であると言える。このフラックス法により成長したGa₂N柱状結晶に関しては、Skrommeらも光学的評価²⁵⁾を実施しており、他の手法で

成長した結晶に比較しても優れた品質であると評価している。

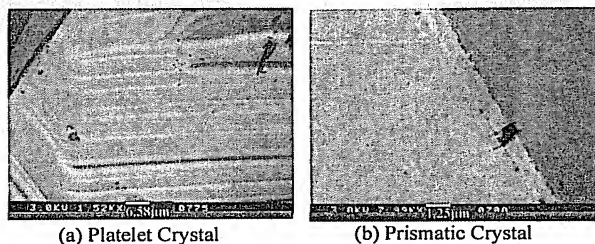


Fig.14 Cathode Luminescence images of platelet and prismatic crystals.

2-4 種結晶成長

2-4-1 これまでの経緯、課題と目的

フラックス法を用いた種結晶成長に関しては、最初に我々のグループ（東北大とリコー）からの報告²⁰⁾がある。前述したFig.6のP-T図の領域②において種結晶成長したGa₂N結晶の顕微鏡写真をFig.15に示す。成長条件は、成長温度850℃、窒素圧力2MPaであり、(a)のP-T図において種結晶成長領域である。(b)は結晶成長前の種結晶、(c)は成長後の結晶、(d)は(c)の断面写真である。この結果から種結晶上に結晶成長していることが判る。しかし種結晶成長はしているものの、この結晶のX線ロッキングカーブの半値全幅は486arcsec.と種結晶の数十arcsec.と比較して結晶品質は悪化している。また種結晶は無色透明であったが、種結晶成長後結晶は黒色であった。これは窒素圧力を下げたことによる、窒素欠損の増大、あるいは窒素サイトへの不純物の置換等が考えられる。

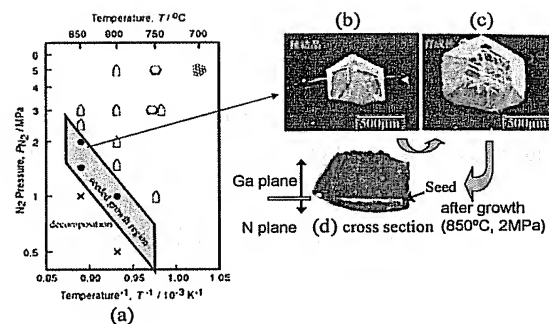


Fig.15 P-T diagram and photographs of GaN crystals for seed growth region. (a) P-T diagram, (b) crystal before seed growth, (c) crystal after seed growth and (d) cross-section of crystal(c).

その後、F. Kawamuraらは種結晶として、サファイア基板上Ga₂N薄膜を用いて、その基板上にフラックス法Ga₂N結晶を成長させることで高品質化の可能性を示した²⁶⁾。用いた基板であるGa₂N薄膜のEPD $1.3 \times 10^6 \text{ cm}^{-2}$ に対して、その上に成長したフラックス法Ga₂N結晶のエッチピットは観察されなかったと報告している。しかしその表面の凹凸は数100 μm ある²⁷⁾。

我々はフラックス法による種結晶成長として、種結晶に対するエピタキシャル成長且つ表面平坦性の向上を目的に新しい種結晶成長方法の研究開発を実施した。

2-4-2 実験方法及び結果

Fig.16に示すように坩堝の底面に種結晶を置き、成長条件として、温度725~825℃、窒素圧力1~5 MPa、Naモル量比 $\text{Na}=\text{Na}/(\text{Na} + \text{Ga})$; 0.3~0.9で結晶成長した。ここで種結晶として、サファイア基板上Ga₂N薄膜（MOVPE法で成長）を用いた。サファイア基板はC面（0001）を用いており、その上に成長したGa₂N結晶もC面である。

キーパラメータを窒素圧力とNaモル量比と捉え、それぞれの表面モフォロジーに対する影響を調べた。

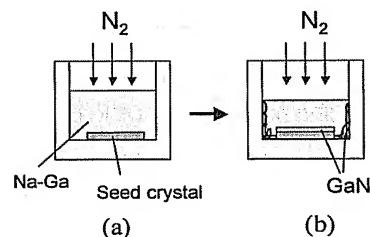


Fig.16 Schematic cross section of crucible for seed growth.

最初に窒素圧力依存性について結果を示す。Fig.17に窒素圧力を0.5～4MPaに変化させ成長したGa₂N結晶のSEM写真を示す。このとき、成長温度775℃とNaモル量比0.67は一定とした。(a), (b), (c), (d)はそれぞれ0.5, 1, 1.5, 4MPaの表面写真であり、(e), (f)はそれぞれ(a), (b)の断面写真である。

(a)と(b)は膜状に成長しており、基板と同じ結晶方位で種結晶成長している。一方(c)と(d)は多核成長しており、基板とは異なる結晶方位で成長している。(a)と(b)は種結晶成長しているものの、断面写真(e)と(f)から判るように、その表面の凹凸は大きい。これは{10-11}の結晶面が成長しているためである。

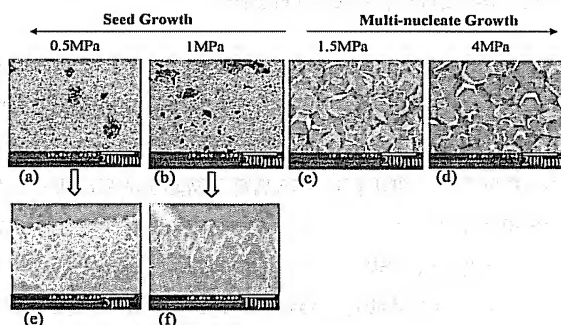


Fig.17 SEM images of GaN crystals at various P_{N_2} . (a), (b), (c) and (d) are 0.5, 1, 1.5 and 4MPa, respectively. (e) and (f) are cross-section of (a) and (b).

次に、表面モフォロジーのNaモル量比依存性について述べる。Fig.18にNaモル量比を0.3～0.67に変化させ成長したGa₂N結晶のSEM写真を示す。このとき、成長温度775℃と窒素圧力4MPaは一定とした。(a), (b), (c)はそれぞれ $r_{Na} = 0.3, 0.4, 0.67$ の表面写真であり、(d), (e)はそれぞれ(a), (b)の断面写真である。

(a)と(b)は膜状に成長しており、基板と同じ結晶方位で種結晶成長している。一方(c)は多核成長しており、基板とは異なる結晶方位で成長している。(a)と(b)は種結晶成長しているものの、断面写真(e)と(f)から判るように、その表面の凹凸は大きく{10-11}の結晶面が成長している。

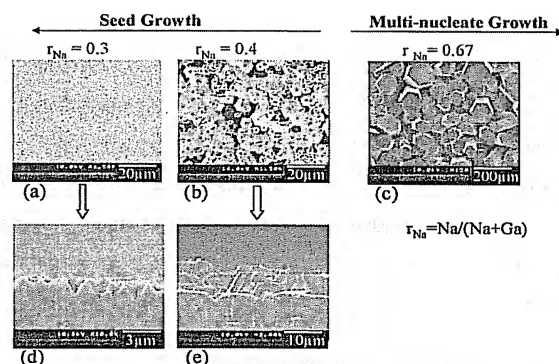


Fig.18 SEM images of GaN crystals at various r_{Na} . (a) $r_{Na}=0.3$, (b) $r_{Na}=0.4$, (c) $r_{Na}=0.67$, (d) and (e) are cross-sectional images of (a) and (b) respectively.

以上のまとめたものをFig.19に示す。横軸に窒素圧力、縦軸にNaモル量比をとると、グラフ右上では多核成長し、グラフ右下では種結晶成長していることが判る。これは、窒素圧力とNa量比が大きくなるとNa-Ga混合融液への窒素溶解度が高くなり、混合融液中のGa₂N過飽和度が高くなることにより、多核成長し易くなり、逆にこれらのパラメータが小さくなると種結晶成長し易くなるものと考えられる。

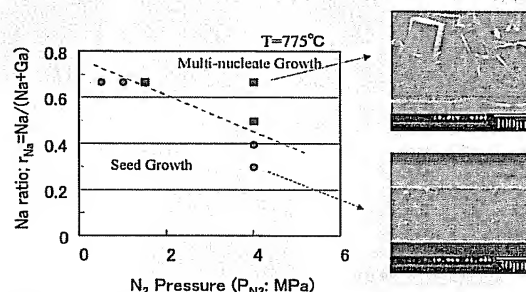


Fig.19 Effect of P_{N_2} and r_{Na} on crystal morphology.

このように基板と同じ結晶方位で種結晶成長はしているものの、{10-11}面が成長することにより、表面の平坦性は悪いという課題が残る。基板と同じ結晶方位で種結晶成長させるためには、窒素の溶解度が小さくなる条件にすれば良いが、Fig.10で前述したように窒素の溶解度が小さくなると{10-11}の結晶方位面が成長し易くなる。逆にC面(0001)が支配的な成長モードにするためには、窒素の溶解度を大きくする必要があるが、その場合には多核成長してしまう。即ち基板と同じ結晶方位での種結晶成長と表面の平坦性はトレード

ドオフの関係にある。

我々のグループではこれまで、フラックス法の自発核成長において添加剤を用いることで成長する結晶方位の異方性を変えることが可能であることを見出していた^{21, 28)}。この中でLiを添加することによりC面が成長し易くなることが判っていた。そこで種結晶成長にもこれを応用展開し、Li添加により基板と同じ結晶方位での種結晶成長と表面平坦性の両立を図ることとした。

2-4-3 Li添加による表面平坦性の改善

Fig.18の(b)の成長条件 ($r_{Na} = 0.4$, $T = 775^{\circ}\text{C}$, $P_{N_2} = 4\text{MPa}$) に対してLiを添加した。Liは Li_3N として添加している。Fig.20に表面SEM像としての結果を示す。(a), (b), (c), (d)はそれぞれ Li_3N 添加量が0, 0.1, 0.13, 0.2mmolに対応している。(a)に対して、Liを添加した(b)と(c)は表面平坦性が改善しているが、添加量を多くした(d)は多核成長していることが判る。(e)と(f)はそれぞれ(a)と(c)の拡大したものである。(a), (b)及び(c)は基板と同じ結晶方位で成長する種結晶成長となっている。

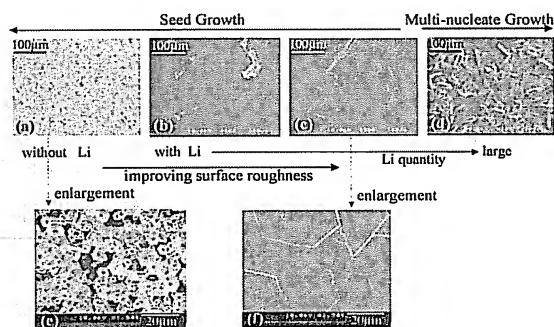


Fig.20 SEM images of GaN crystals at various quantities of Li additive. Quantities of Li are (a) 0, (b) 0.1, (c) 0.13 and (d) 0.2mmol respectively. (e) and (f) are enlargements of (a) and (c).

Fig.21にFig.20 (c)の断面SEM像とそれから評価した表面荒さを示す。(a)はLi添加無し、(b)はLi添加したものである。Li添加無しでは $\{10\cdot11\}$ 面が成長し、表面の凹凸は $10\sim20\mu\text{m}$ と大きい、Li添加することによりC面(0001)面が支配的な成長となり、表面平坦性を大幅に改善することが出来た。

Li添加により成長速度の結晶方位異方性が高まり、C面が支配的な成長が促進されたと考えられる。

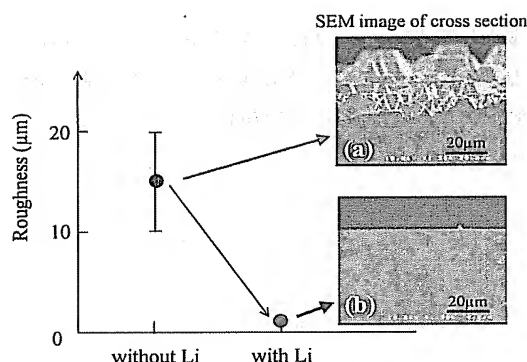


Fig.21 Surface roughness on GaN crystals without and with Li additives.

3. まとめと今後の展開

フラックス法によるGaN結晶成長について、その特徴を述べ、これまでの技術の進展を我々の実験結果も含めて説明した。フラックス法は原理的に溶液成長の一種であり、他の手法と比較しても低温低压で高品質な結晶が成長出来た。外部窒素導入法や加熱方式等の方式や装置上の工夫により、結晶サイズの拡大を実現した。窒素圧力と成長温度を制御することにより、自発核成長と種結晶成長の成長モード制御及び自発核成長での結晶形態制御を可能とした。自発核成長で得られた結晶の品質は、これまで高品質と言われてきた高压溶液法と同等あるいはそれ以上であり、特に柱状結晶の結晶性は優れていることが判った。種結晶成長ではLi添加によりこれまで困難であった、表面モフォロジーの改善を実現することが出来た。

今後は自発核成長での更なる結晶サイズ拡大を図り、デバイス応用可能な高品質GaN結晶の実現を目指すと共に、種結晶成長での結晶品質の定量的評価によりその技術的ポテンシャルを明らかにしていく。

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(12) United States Patent
Kitaoka et al.**(10) Patent No.: US 7,524,691 B2**
(45) Date of Patent: Apr. 28, 2009**(54) METHOD OF MANUFACTURING GROUP III NITRIDE SUBSTRATE**

6,503,610 B2 1/2003 Hiramatsu et al.

(75) Inventors: Yasuo Kitaoka, Ibaraki (JP); Hisashi Minemoto, Hirakata (JP); Isao Kidoguchi, Kawanishi (JP); Akihiko Ishibashi, Mishima-gun (JP)

(Continued)

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(73) Assignee: Panasonic Corporation, Osaka (JP)**(*) Notice:** Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 614 days.

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(22) Filed: Jan. 16, 2004

(Continued)

(65) Prior Publication Data

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(30) Foreign Application Priority Data

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Primary Examiner—W. David Coleman
Assistant Examiner—Sonya D McCall Shepard
(74) Attorney, Agent, or Firm—Hamre, Schumann, Mueller & Larson, P.C.**(57) ABSTRACT****(51) Int. Cl.**
H01L 21/00 (2006.01)**(52) U.S. Cl.** 438/46; 257/90**(58) Field of Classification Search** 438/46,
438/47, 22, 478

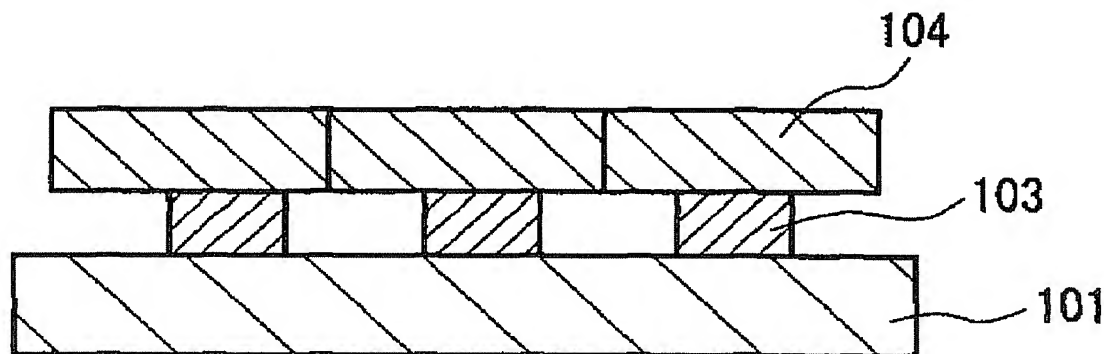
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The present invention provides a manufacturing method that makes it possible to manufacture a substrate that is formed of high-quality Group III nitride crystals alone and has less warping. A Group III nitride layer (a seed layer and a selective growth layer) including gaps is formed on a substrate (a sapphire substrate). In an atmosphere containing nitrogen, the surface of the Group III nitride layer is brought into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, and thereby the at least one Group III element and the nitrogen are made to react with each other to grow Group III nitride crystals (GaN crystals) on the Group III nitride layer. Thereafter, a part including the substrate and a part including the Group III nitride crystals are separated from each other in the vicinities of the gaps.

25 Claims, 11 Drawing Sheets

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FIG. 1A

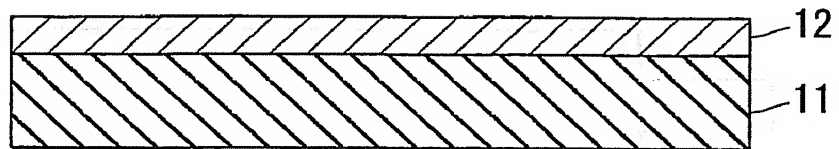


FIG. 1B

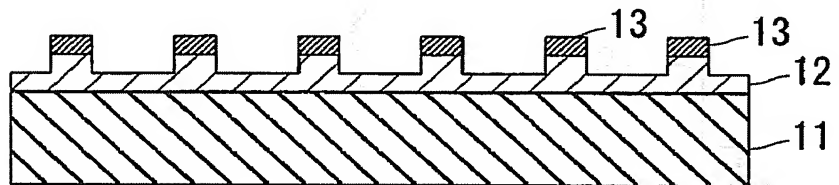


FIG. 1C

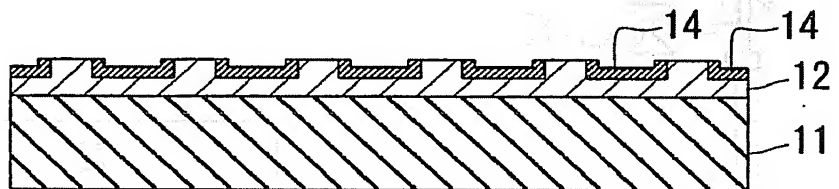


FIG. 1D

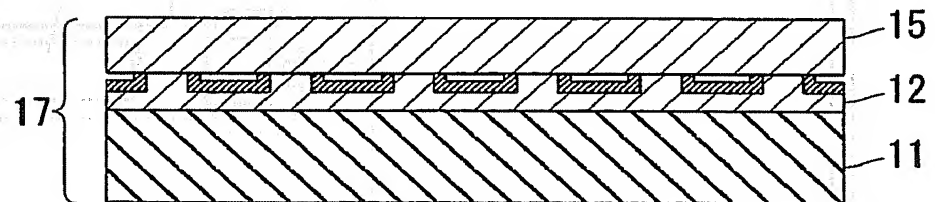


FIG. 1E

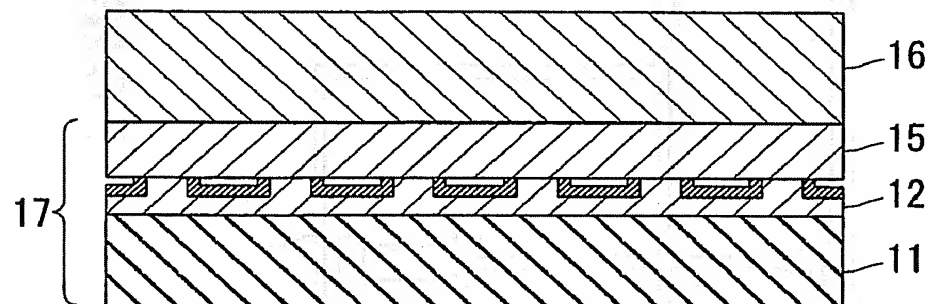
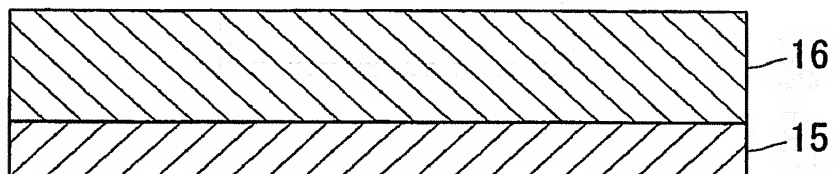
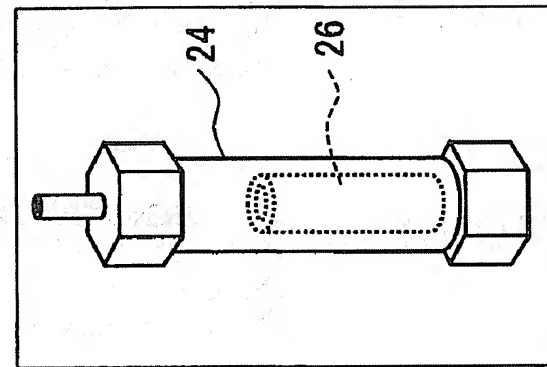
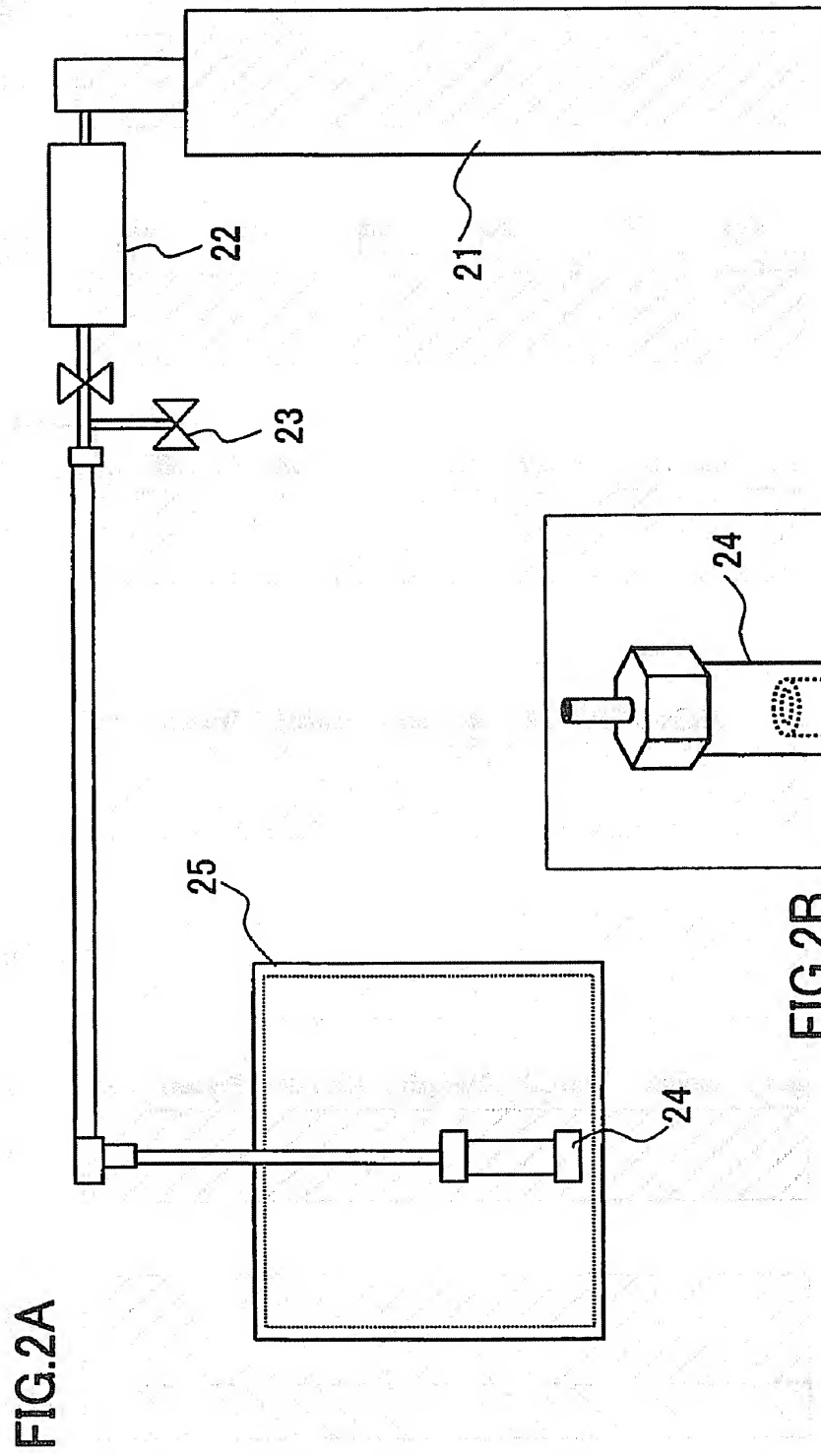


FIG. 1F





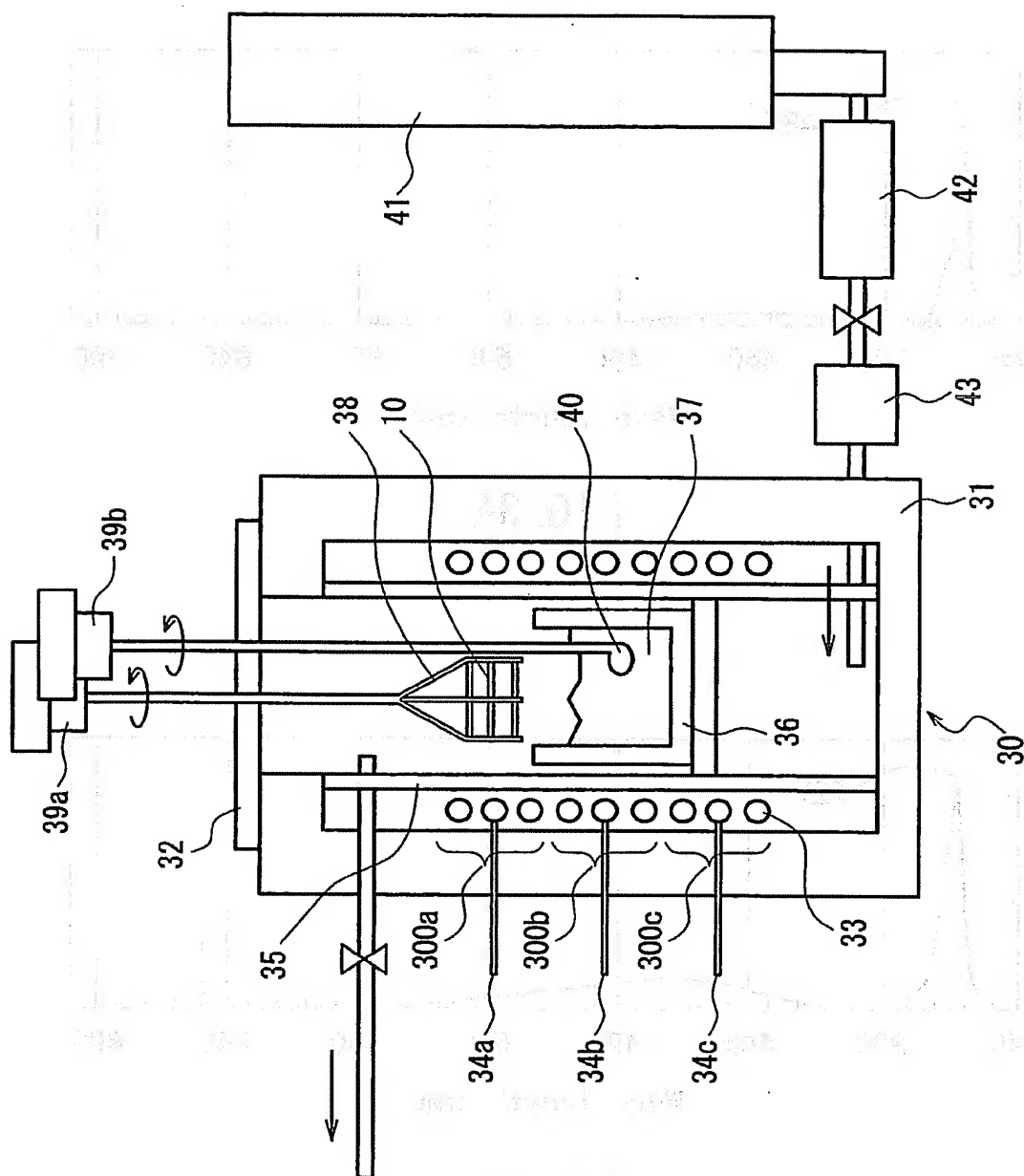


FIG. 4

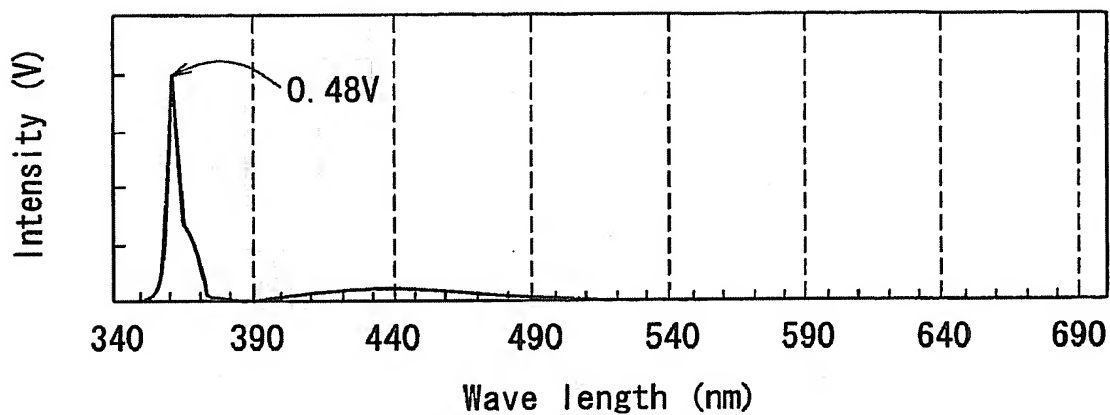


FIG. 3A

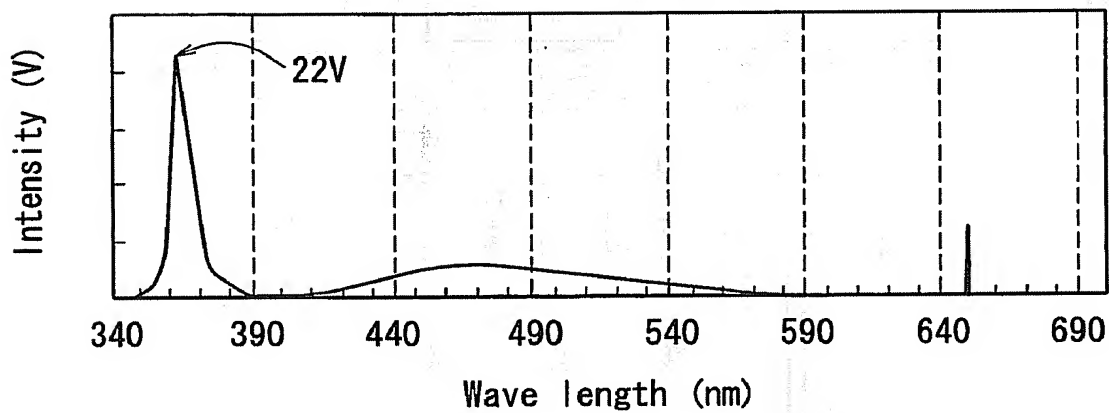


FIG. 3B

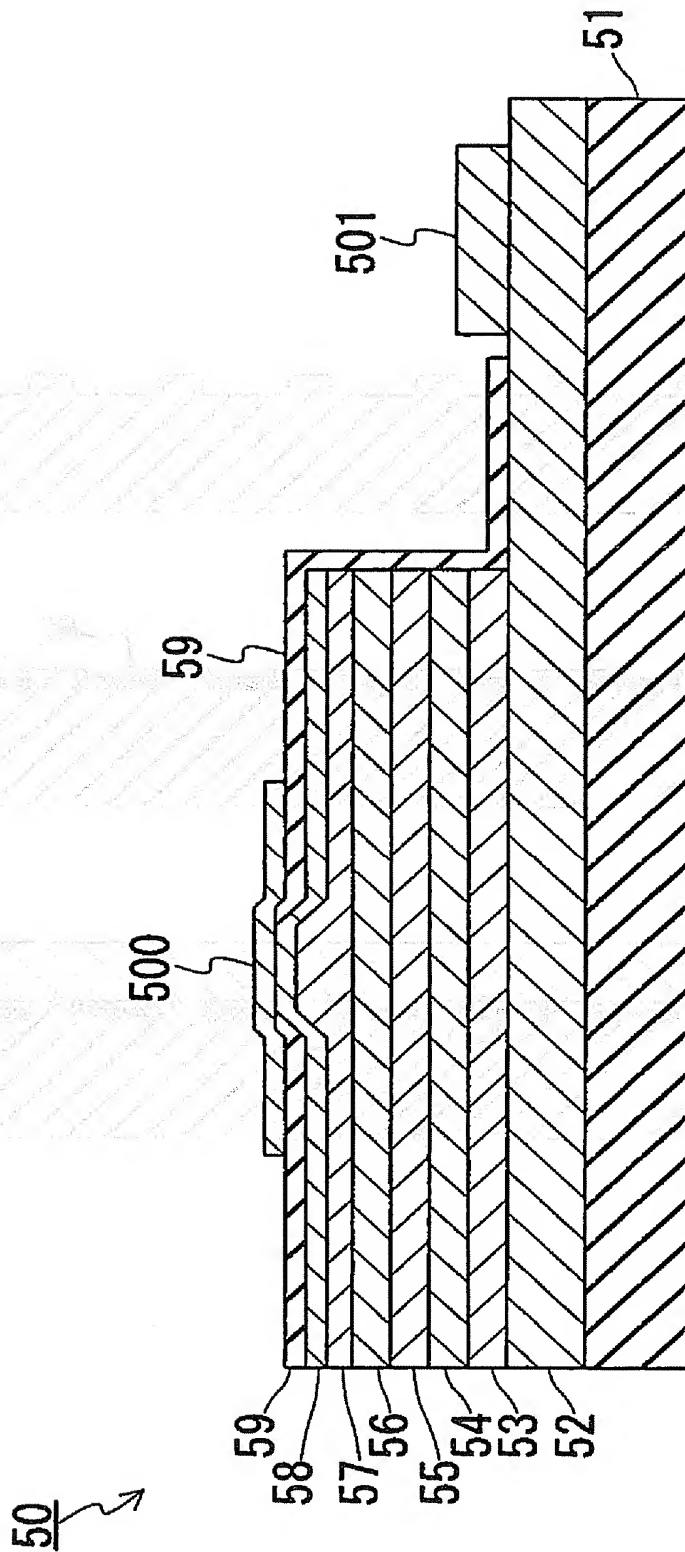


FIG.5

FIG. 6A

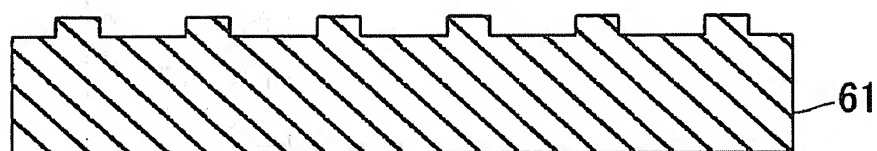


FIG. 6B

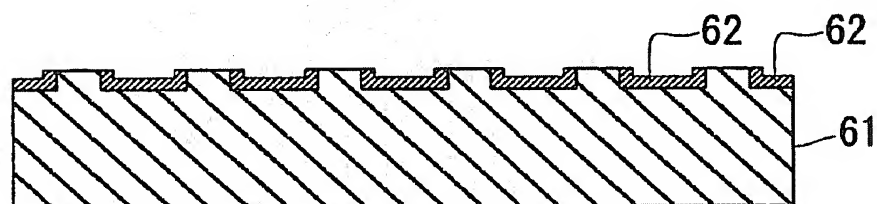
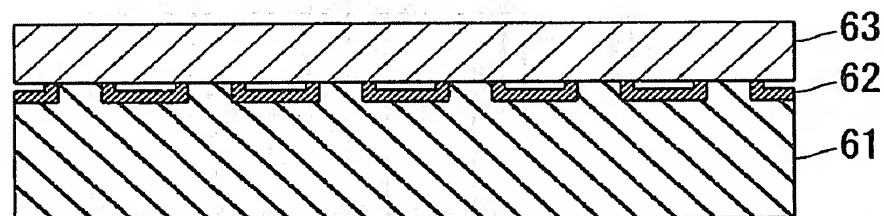


FIG. 6C



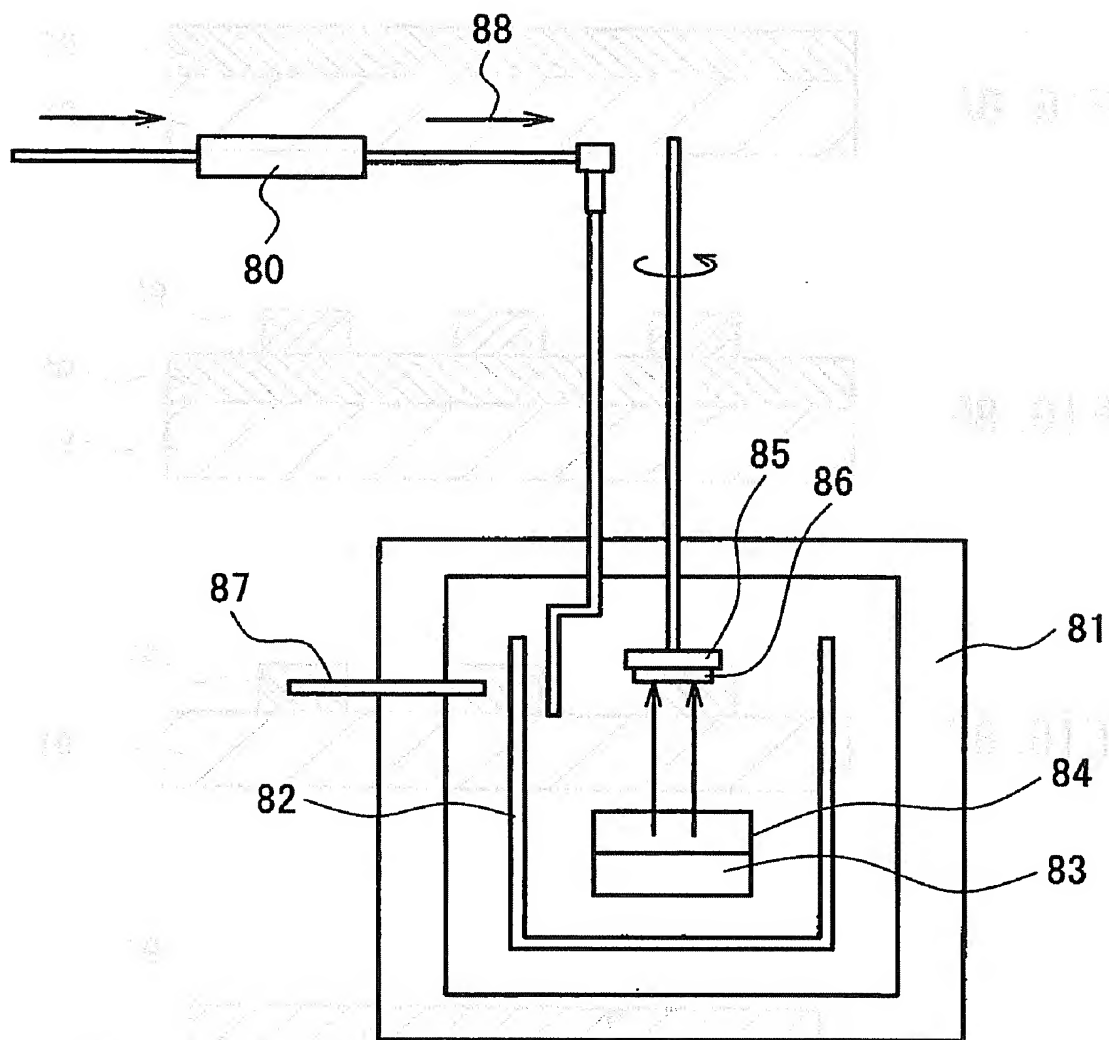


FIG. 7

FIG. 8A

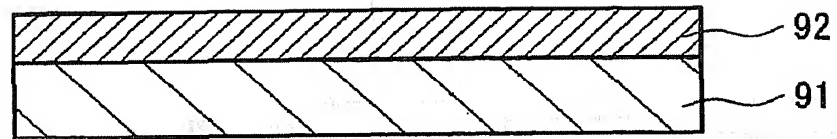


FIG. 8B

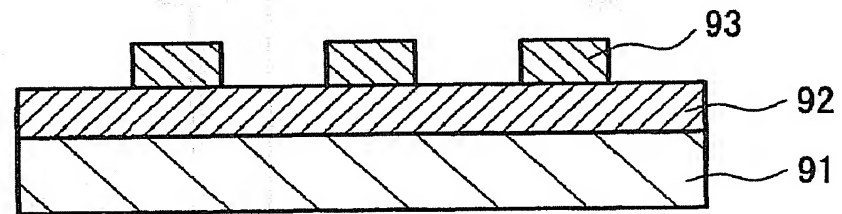


FIG. 8C

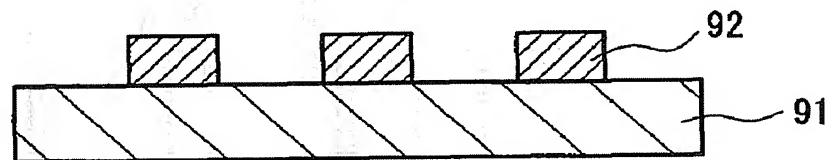


FIG. 8D

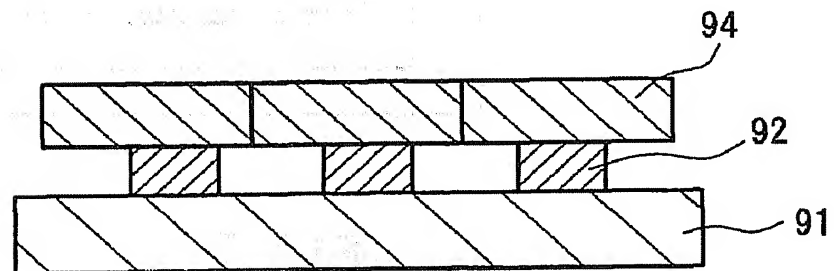


FIG. 9A

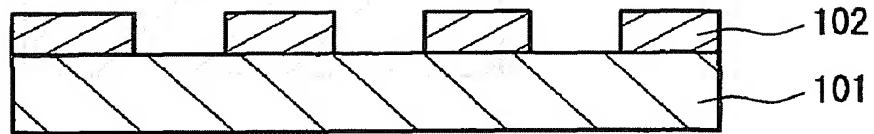


FIG. 9B

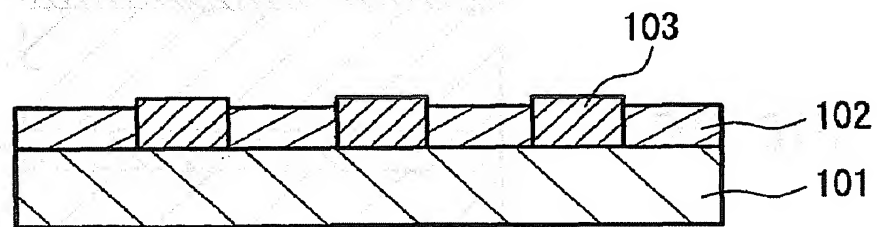


FIG. 9C

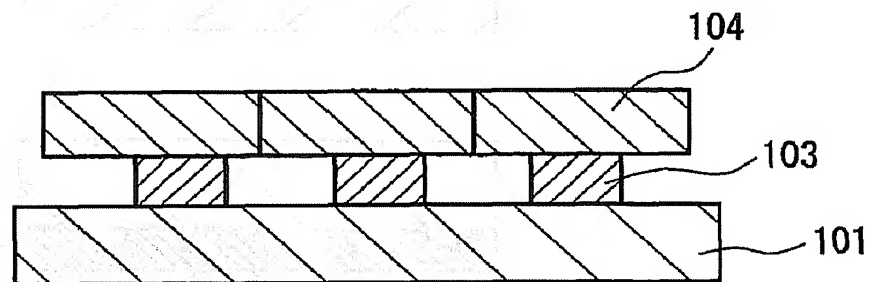


FIG. 10A

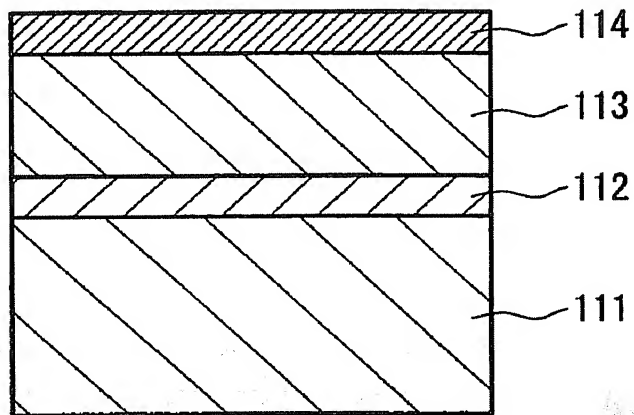


FIG. 10B

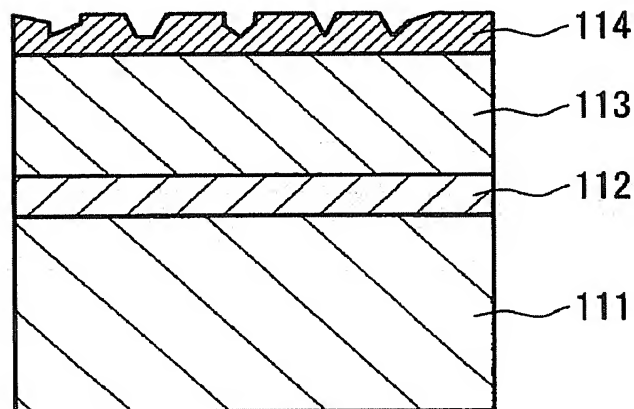


FIG. 10C

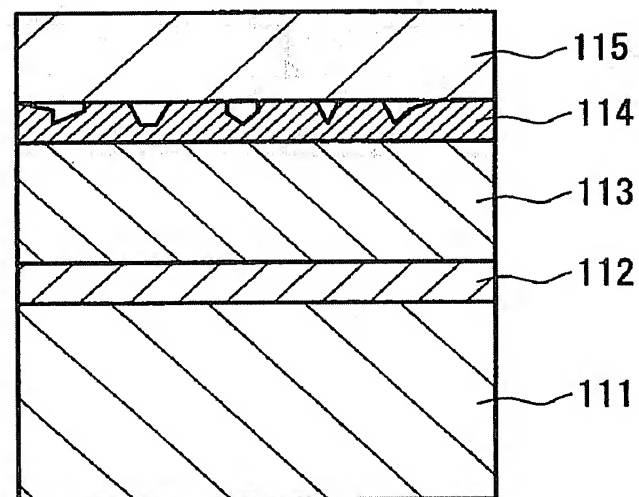


FIG. 11A

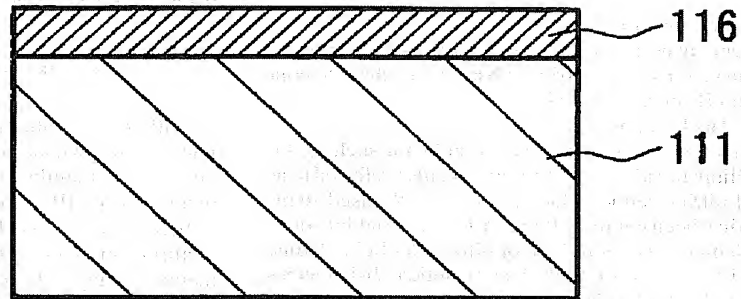


FIG. 11B

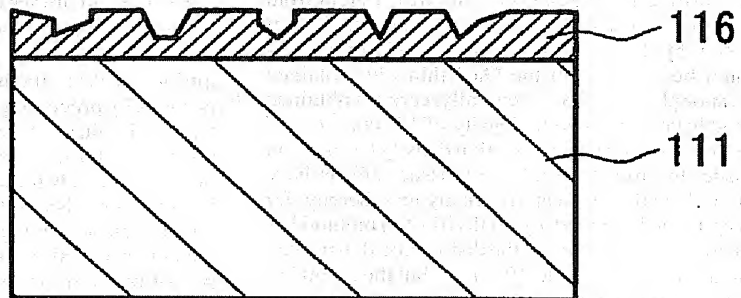
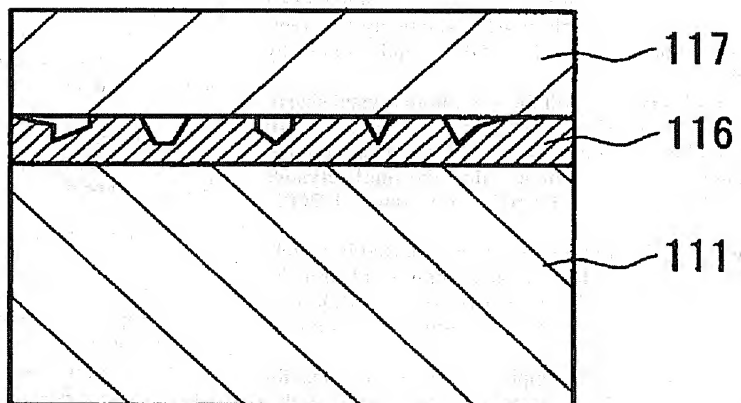


FIG. 11C



METHOD OF MANUFACTURING GROUP III NITRIDE SUBSTRATE

BACKGROUND OF THE INVENTION

1. Field of the Invention

The present invention relates generally to a method of manufacturing a Group III nitride substrate (a substrate containing Group III nitride crystals).

2. Related Background Art

A Group III nitride compound semiconductor such as, for instance, gallium nitride (GaN) (hereinafter also referred to as a "Group III nitride semiconductor" or a "GaN-based semiconductor") has been gaining attention as a material for semiconductor elements that emit blue or ultraviolet light. A blue laser diode (LD) is used for high-density optical disk devices or displays while a blue light emitting diode (LED) is used for displays, lighting, etc. It is expected to use an ultraviolet LD in the field of biotechnology or the like and an ultraviolet LED as an ultraviolet source for a fluorescent lamp.

Generally, substrates of a Group III nitride semiconductor (for example, GaN) that are used for LDs or LEDs are formed through vapor phase epitaxy. For instance, substrates have been used that are obtained through heteroepitaxial growth of Group III nitride crystals on a sapphire substrate. However, the sapphire substrate and the GaN crystals are different from each other in lattice constant by 13.8% and in coefficient of linear expansion by 25.8%. Hence, a sufficiently high crystallinity cannot be obtained in the GaN thin film obtained through the vapor phase epitaxy. Generally, crystals obtained by this method have a dislocation density of 10^8 cm^{-2} to 10^9 cm^{-2} and thus the reduction in dislocation density has been an important issue. In order to resolve this issue, efforts have been made to reduce the dislocation density and thereby, for example, an epitaxial lateral overgrowth (ELOG) method has been developed. With this method, the dislocation density can be reduced to around 10^5 cm^{-2} to 10^6 cm^{-2} , but the manufacturing process is complicated.

On the other hand, besides the vapor phase epitaxy, a method of carrying out crystal growth from liquid phase also has been studied. However, since the equilibrium vapor pressure of nitrogen is at least 10000 atm at the melting point of a Group III nitride single crystal such as, for instance, GaN or AlN, conventionally it has been understood that a condition of 8000 atm at 1200° C. is required for growing GaN from a liquid phase. In this connection, recently, it was made clear that GaN was able to be synthesized at relatively low temperature and pressure, specifically, 750° C. and 50 atm, by using a Na flux.

Recently, single crystals with the maximum crystal size of about 1.2 mm are obtained by a method in which a mixture of Ga and Na is melted in a nitrogen gas atmosphere containing ammonia at 800° C. and 50 atm, and then the single crystals are grown for 96 hours using the melt (for instance, JP2002-293696A).

Furthermore, another method has been reported in which, after a GaN crystal layer is formed on a sapphire substrate by a metal organic chemical vapor deposition (MOCVD) method, single crystals are grown by a liquid phase epitaxy (LPE) method.

Generally, for instance, a sapphire substrate is used for manufacturing a Group III nitride substrate. However, such a substrate and a Group III nitride crystal are different from each other in lattice constant and coefficient of thermal expansion. Hence, when Group III nitride crystals are grown using such a substrate, the substrate may be distorted or warped in some cases. Accordingly, the substrate may be

damaged during crystal growth or it may be difficult to manufacture a device using a semiconductor substrate formed in the above-mentioned manner in some cases. For instance, mask alignment may be difficult, for example in the case of using a stepper to be employed in a device manufacturing process.

SUMMARY OF THE INVENTION

With such a situation in mind, the present invention is intended to provide a manufacturing method that makes it possible to manufacture a substrate that is formed of high-quality Group III nitride crystals alone and has less warping.

In order to achieve the aforementioned object, a first manufacturing method of the present invention is a method of manufacturing a Group III nitride substrate including the steps of (i) forming a Group III nitride layer including gaps, on a substrate; (ii) bringing a surface of the Group III nitride layer into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen, to make the at least one Group III element and the nitrogen react with each other to grow Group III nitride crystals on the Group III nitride layer; and (iii) separating a part including the substrate and a part including the Group III nitride crystals from each other in the vicinities of the gaps.

Furthermore, a second manufacturing method of the present invention is a method of manufacturing a Group III nitride substrate including the steps of (I) forming convex portions by processing a surface of a substrate; (II) growing a Group III nitride layer from upper surfaces of the convex portions to form a seed crystal substrate having gaps formed between the substrate and the Group III nitride layer; (III) bringing a surface of the Group III nitride layer into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in a pressurized atmosphere containing nitrogen, to make the at least one Group III element and the nitrogen react with each other to grow Group III nitride crystals on the Group III nitride layer; and (IV) separating a part including the substrate and a part including the Group III nitride crystals from each other in the vicinities of the gaps.

In this specification, the "Group III nitride" denotes a semiconductor expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$) unless otherwise specified. In this connection, it is understood that a formula of $0 \leq 1-x-y \leq 1$ is satisfied since the composition ratio can never be a negative value (the same applies to other composition formulae).

According to the manufacturing methods of the present invention, a substrate can be manufactured easily that is formed of high-quality Group III nitride crystals alone and has less warping.

BRIEF DESCRIPTION OF THE DRAWINGS

FIGS. 1A to 1F are cross-sectional views showing steps of an example of a manufacturing method according to the present invention.

FIGS. 2A and 2B are schematic views showing an example of a manufacturing apparatus used in manufacturing methods of the present invention.

FIG. 3A is a graph showing PL intensity of a GaN crystal obtained by a conventional method; and FIG. 3B is a graph showing PL intensity of a GaN crystal obtained in one example of the present invention.

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FIG. 4 is a schematic view showing another example of a manufacturing apparatus used in the manufacturing method of the present invention.

FIG. 5 is a cross-sectional view showing an example of a semiconductor device including a substrate manufactured by a manufacturing method of the present invention.

FIGS. 6A to 6C are cross-sectional views showing steps of another example of a manufacturing method according to the present invention.

FIG. 7 is a schematic view showing still another example of a manufacturing apparatus used in a manufacturing method of the present invention.

FIGS. 8A to 8D are cross-sectional views showing steps of a further example of a manufacturing method according to the present invention.

FIGS. 9A to 9C are cross-sectional views showing steps of still another example of a manufacturing method according to the present invention.

FIGS. 10A to 10C are cross-sectional views showing steps of yet another example of a manufacturing method according to the present invention.

FIGS. 11A to 11C are cross-sectional views showing steps of a further example of a manufacturing method according to the present invention.

DETAILED DESCRIPTION OF THE INVENTION

Hereinafter, the present invention is described in detail.

In the manufacturing methods of the present invention, it is preferable that the at least one Group III element is gallium, and the Group III nitride crystals are GaN crystals.

In the manufacturing methods of the present invention, it is preferable that the atmosphere containing nitrogen is a pressurized atmosphere. The atmosphere is pressurized, for example, in a range of 0.1 MPa to 10 MPa, and preferably in a range of 0.5 MPa to 5 MPa.

It is preferable that in the step (iii) of the first manufacturing method of the present invention, the separation is carried out using stress generated by a difference in coefficient of linear expansion between the substrate and the Group III nitride crystals.

In the first manufacturing method of the present invention, it is preferable that the step (i) includes the steps of: (i-1) forming a first semiconductor layer expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$), on the substrate; (i-2) forming convex portions by partially removing the first semiconductor layer; and (i-3) forming the Group III nitride layer having gaps in its portions other than the convex portions by growing a second semiconductor layer from upper surfaces of the convex portions of the first semiconductor layer, with the second semiconductor layer being expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$), wherein in the step (iii), the first semiconductor layer and the second semiconductor layer are separated from each other at the upper surfaces of the convex portions.

Preferably, the upper surfaces are C-planes.

It is preferable that in the step (i-2), the convex portions are formed in stripes.

It is preferable that in the step (i-2), concave portions that are portions other than the upper surfaces of the convex portions are covered with a mask film. Preferable materials used for forming the mask film include silicon nitride, oxide silicon, nitride oxide silicon, aluminum oxide, aluminum nitride oxide, titanium oxide, zirconium oxide, niobium oxide, tungsten, molybdenum, niobium, tungsten silicide, molybdenum silicide, and niobium silicide. They may be used individually,

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or two or more of them may be used together. It is preferable that the mask film is formed of high melting metal or a high melting metallized material.

In the manufacturing methods of the present invention, it is preferable that the substrate is a sapphire substrate.

In the manufacturing methods of the present invention, it is preferable that the alkali metal is at least one selected from sodium, lithium, and potassium.

In the manufacturing methods of the present invention, it is preferable that the melt further contains alkaline-earth metal.

In the first manufacturing method of the present invention, it is preferable that the step (i) includes the steps of: (i-a) forming a first semiconductor layer expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$), on the substrate; and (i-b) forming concave portions to be gaps by partially removing the first semiconductor layer to expose portions of the substrate and thereby forming remaining portions into convex portions, wherein in the step (ii), Group III nitride crystals are grown on surfaces of the convex portions formed in the step (i-b).

In the first manufacturing method of the present invention, it is preferable that the step (i) includes the steps of (i-c) forming a mask film patterned on the substrate; and (i-d) forming a first semiconductor layer having a convex shape on each of portions of the substrate that are not covered with the mask film, to allow concave portions to be gaps. The first semiconductor layer can be expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$). The concave portions are portions where the first semiconductor layer having the convex shape has not been formed. In the step (ii), the Group III nitride crystals are grown on a surface of the first semiconductor layer formed in the step (i-d).

In the first manufacturing method of the present invention, it is preferable that in the step (i), the Group III nitride layer including gaps includes a semiconductor layer expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$). After forming the semiconductor layer, the gaps are formed in the semiconductor layer or at a surface of the semiconductor layer through a temperature-programmed heat treatment carried out in an atmosphere of a mixture of ammonia and nitrogen. In this case, it is preferable that the Group III nitride layer including gaps is a semiconductor layer expressed by a composition formula of $\text{Ga}_x\text{In}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$). Preferably, the temperature-programmed heat treatment is carried out at a programming rate of 50 to 100° C./min.

In the manufacturing methods of the present invention, it is preferable that a cycle of the gaps is at least 30 μm , more preferably at least 50 μm , and further preferably 100 μm . The "gaps" denote spaces observed at a crystal cross section and are spaces (for instance, on the order of sub microns) that can be observed with a scanning electron microscope (SEM) or larger spaces than those. Furthermore, the "cycle of gaps" denotes a distance or a pitch between two gaps adjoining each other when a plurality of gaps are present. This can be measured with, for example, an electron microscope. The "cycle of gaps" also can be expressed by, for instance, a distance (a pitch) between growing points, at the surfaces of the convex portions or the like, from which Group III nitride crystals grow. Furthermore, the cycle of gaps can be adjusted through a suitable selection of a manner of forming the growing points (for example, their size, their positions, or the distance between adjoining two of them).

The Group III nitride substrate of the present invention is a Group III nitride substrate manufactured by a manufacturing method of the present invention.

In the substrate of the present invention, a cycle of dense dislocation areas is preferably at least 30 μm , more preferably at least 50 μm , and further preferably at least 100 μm . The "dense dislocation areas" denote areas where the number of edge dislocations or screw dislocations is at least 10^7 to 10^8 per cm^2 . Examples of the method of determining it include a method in which cathodoluminescence generated by electron beam irradiation is observed and thereby the number of dislocations is determined from the number of dark spots, and a method in which etching is carried out with acid (200° C.) such as, for example, pyrophosphoric acid, and then concavities and convexities are observed by atomic force microscopy (AFM) or the like.

In the substrate of the present invention, a dislocation density is preferably at least $1 \times 10^5 \text{ cm}^{-2}$, more preferably at least $1 \times 10^4 \text{ cm}^{-2}$. The method of determining dislocation density is the same as that of determining the dense dislocation areas, for example a method in which cathodoluminescence generated by electron beam irradiation is observed and thereby the number of dislocations is determined from the number of dark spots, and a method in which etching is carried out with acid (200° C.) such as, for example, pyrophosphoric acid, and then concavities and convexities are observed by atomic force microscopy (AFM) or the like.

A semiconductor device of the present invention includes a substrate and a semiconductor element formed on the substrate, wherein the substrate is a Group III nitride substrate manufactured by a manufacturing method of the present invention. Preferably, the semiconductor element is a laser diode or a light emitting diode.

Hereinafter, embodiments of the present invention are described using examples.

A method of the present invention is a method for manufacturing a Group III nitride substrate. According to this method, a substrate can be manufactured that is formed of Group III nitride semiconductor single crystals alone.

In this method, first, a Group III nitride layer including gaps is formed on a substrate (step (i)). The substrate can be, for example, a sapphire substrate. An example of a method of forming the Group III nitride layer including gaps is described below.

First of all, a first semiconductor layer expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$) is formed on the substrate (step (i-1)). The first semiconductor layer can be formed by the MOCVD method or a molecular beam epitaxy (MBE) method.

Subsequently, the first semiconductor layer is removed partially to form convex portions (step (i-2)). The convex portions can be formed by a well-known method using photolithography and etching in combination. Usually, the upper surfaces of the convex portions are C-planes. The shape of the convex portions can be a shape that facilitates the separation of the substrate in the step carried out later. The convex portions may be formed, for instance, in stripes or dots. It is preferable that the area of the convex portions does not exceed 50% of the whole area. When being formed in stripes, the convex portions each have a width of, for example, 1 μm to 5 μm , and an interval between two adjacent convex portions is, for instance, 5 μm to 20 μm .

In the step (i-2), it is preferable that portions (concave portions) that are those other than the upper surfaces of the convex portions are covered with a mask film. The mask film can be formed of, for instance, silicon nitride, oxide silicon, nitride oxide silicon, aluminum oxide, aluminum nitride oxide, titanium oxide, zirconium oxide, or niobium oxide. The mask film also may be formed of high melting metal or a high melting metallized material that each have a high melt-

ing point (a melting point of at least 1000° C.). Specifically, the mask film may be formed of tungsten, molybdenum, niobium, tungsten silicide, molybdenum silicide, or niobium silicide.

Method of Forming Seed Layer including Gap Layer

Next, the Group III nitride layer having gaps in its portions other than the convex portions is formed by growing a second semiconductor layer from upper surfaces of the convex portions of the first semiconductor layer, with the second semiconductor layer being expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$) (step (i-3)). The second semiconductor layer may have the same composition as that of the first semiconductor layer or may have a different composition. The second semiconductor layer can be formed by, for instance, the MOCVD method or the MBE method. The second semiconductor layer that is grown from the upper surfaces (C-planes) of the convex portions grows upward as well as in the lateral direction. Accordingly, portions of the second semiconductor layer that have grown from convex portions adjoining each other dock with each other spatially and thereby a bridge structure is formed. In this manner, the Group III nitride layer is formed with gaps formed in its portions other than the convex portions. Using a crystal layer of the Group III nitride semiconductor thus obtained as a seed crystal, crystal growth is carried out in the following step (ii).

Next, in an atmosphere containing nitrogen (preferably, a pressurized atmosphere having 100 atm or lower), a surface of the Group III nitride layer is brought into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, to make the at least one Group III element and the nitrogen react with each other and thereby Group III nitride crystals are grown on the Group III nitride layer (step (ii)). The atmosphere containing nitrogen can be, for example, nitrogen gas or a nitrogen gas atmosphere containing ammonia. The alkali metal can be at least one selected from sodium, lithium, and potassium, i.e. one of them or a mixture thereof, which usually is used in a flux state.

The melt is prepared, for instance, by placing a material in a crucible and heating it. After the preparation of the melt, the melt is brought into a supersaturation state and thereby semiconductor crystals grow. Melting of the material and crystal growth are carried out at a temperature of about 700° C. to 1100° C. under a pressure of about 1 atm to 50 atm. The melt further may contain alkaline-earth metal. The alkaline-earth metal can be, for example, Ca, Mg, Sr, Ba, or Be.

According to this method, Group III nitride crystals can be obtained that are expressed by a composition formula of $\text{Al}_s\text{Ga}_t\text{In}_{1-s-t}\text{N}$ (wherein $0 \leq s \leq 1$ and $0 \leq t \leq 1$). For instance, when using gallium alone as the Group III element of the material, GaN crystals are obtained, while when using gallium and aluminum as the Group III element of the material, crystals are obtained that are expressed by a composition formula of $\text{Al}_s\text{Ga}_{1-s}\text{N}$ (wherein $0 \leq s \leq 1$).

Next, a part including the substrate and a part including the Group III nitride crystals are separated from each other in the vicinities of the gaps (step (iii)). This separation step may be carried out mechanically or using stress generated by a difference in coefficient of linear expansion between the substrate and the Group III nitride crystals. When utilizing the difference in coefficient of linear expansion, the separation can be carried out, for example, in a cooling step (including natural cooling) following the step (ii). When the steps (i-1) to (i-3) described above are carried out, the first semiconductor

layer and the second semiconductor layer are separated at the upper surfaces of the convex portions in the step (iii).

In this manner, a substrate formed of Group III nitride crystals alone can be obtained. The example described above was directed to the case where the convex portions are formed in the Group III nitride layer. However, the convex portions may be formed in the substrate. A manufacturing method employed in this case is described below.

In this case, first, a surface of a substrate is processed to form convex portions (step (I)). The substrate can be, for instance, a sapphire substrate whose surface is a C-plane. The concave and convex portions can be formed by, for instance, a photolithography process and dry etching using chlorine gas.

Next, a Group III nitride layer is grown from upper surfaces of the convex portions to form a seed crystal substrate having gaps formed between the substrate and the Group III nitride layer (step (II)). The Group III nitride layer can be formed by the MOCVD method or the MBE method. Using a crystal layer of the Group III nitride semiconductor formed in this step as a seed crystal, crystal growth is carried out in the following steps.

Next, in an atmosphere containing nitrogen, a surface of the Group III nitride layer is brought into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, to make the at least one Group III element and the nitrogen react with each other. As a result Group III nitride crystals are grown on the Group III nitride layer (step (III)). Thereafter, a part including the substrate and a part including the Group III nitride crystals are separated from each other in the vicinities of the gaps (step (IV)). Since the steps (III) and (IV) are identical to the steps (ii) and (iii) described above, the descriptions applied to the both are not repeated. In the step (IV), the separation of the substrate is carried out at the upper surfaces of the convex portions of the substrate.

By using a substrate including gaps as a seed crystal, a sapphire substrate can be separated from GaN single crystals. This separation step may be carried out mechanically or using stress generated by a difference in coefficient of linear expansion between the substrate and the Group III nitride crystals. When using the difference in coefficient of linear expansion, the separation can be carried out, for example, in a cooling step (including natural cooling) following the step (III).

In the above-mentioned example, the description was directed to the method in which using a substrate with a Group III nitride layer including gaps formed therein, Group III nitride crystals grown from liquid phase are separated in the vicinities of the gaps. However, the following method also permits the formation of gaps and the separation of Group III nitride crystals.

First, a first semiconductor layer expressed by a composition formula of $\text{Al}_u\text{Ga}_{1-u}\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$) is formed on a substrate (step (i-a)). The first semiconductor layer can be formed by, for instance, the MOCVD method or the MBE method.

Next, the first semiconductor layer is removed partially to expose portions of the upper surface of the substrate and thereby convex portions are formed (step (i-b)). The convex portions can be formed by a well-know method using photolithography and etching in combination. Usually, the upper surfaces of the convex portions are C-planes. The shape of the convex portions can be a shape that facilitates the separation of the substrate in the step carried out later. The convex portions may be formed, for instance, in stripes or dots.

In the present example, in an atmosphere containing nitrogen (preferably, a pressurized atmosphere of 100 atm or

lower), the surface of the first semiconductor layer is brought into contact with a melt containing alkali metal, the nitrogen and at least one Group III element selected from gallium, aluminum, and indium, to make the at least one Group III element and the nitrogen react with each other. As a result Group III nitride crystals are grown on the upper surfaces of the convex portions of the first semiconductor layer (step (ii)).

In this manner, gaps are formed between the substrate and the Group III nitride crystals. In the present example, since the Group III nitride crystals are grown from liquid phase in the melt containing alkali metal, a Group III element and nitrogen, the lateral growth rate can be increased as compared to the case of vapor growth methods such as a conventional metal organic chemical vapor deposition (MOCVD) method and hydride vapor phase epitaxy (HVPE) method. Accordingly, it is preferable that the area of the convex portions does not exceed 10% of the whole area. When being formed in stripes, the convex portions each have a width of, for example, 1 μm to 5 μm , and an interval between two adjacent convex portions is, for instance, 20 μm to 500 μm . A cycle of the convex portions is desirably at least 30 μm , more desirably at least 50 μm , and further desirably at least 100 μm .

In general vapor growth methods such as the MOCVD method, the MBE method, the HVPE method, etc, since the lateral growth rate is low, docking of crystals growing from the respective convex portions with each other as shown in FIG. 1D is not possible unless the interval between convex portions adjoining each other is set at 20 μm or shorter. Furthermore, the convex portions each is required to have several micrometers to make the crystals grow. Hence, the ratio of a region where the convex portions are formed to a region where the convex portions are not formed is about 20% or higher. In the liquid phase growth method, however, the lateral growth can be promoted and larger single crystals can be made to grow from the convex portions. Hence, the ratio further can be reduced. The reduction in the ratio of the convex portions means a reduction in area where the substrate and crystal portions grown epitaxially from the liquid phase are in contact with each other. Accordingly, the substrate can be separated readily in the vicinities of the gaps. Hence, this provides great practical effects. The cycle of dense dislocation areas is desirably at least 30 μm , more desirably at least 50 μm , and further desirably at least 100 μm .

Next, the substrate and the Group III nitride crystals are separated from each other in the vicinities of the gaps (step (iii)). This separation step may be carried out mechanically or using stress generated by a difference in coefficient of linear expansion between the substrate and the Group III nitride crystals. When using the difference in coefficient of linear expansion, the separation can be carried out, for example, in a cooling step (including natural cooling) following the step (ii).

Moreover, another method described below may be used as the method of forming a semiconductor layer to serve as convex portions on the substrate.

First, a mask film is patterned on a substrate, such as a sapphire substrate (step (i-c)). The mask film can be formed of, for example, silicon nitride, oxide silicon, nitride oxide silicon, aluminum oxide, or aluminum nitride oxide. Furthermore, the mask film may be formed of high melting metal or a high melting metallized material that has a high melting point (a melting point of at least 1000° C.). Specifically, the mask film may be formed of titanium, tungsten, molybdenum, niobium, tungsten silicide, molybdenum silicide, or niobium silicide.

Next, a semiconductor layer expressed by a composition formula of $\text{Al}_u\text{Ga}_{1-u}\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$) is

formed on portions of the substrate that are not covered with the mask film (step (i-d)). The semiconductor layer can be formed by, for instance, the MOCVD method or the HVPE method.

Using the seed substrate described above, Group III nitride crystals are grown in a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen (step (ii)).

Finally, the substrate and the Group III nitride crystals are separated from each other in the vicinities of the gaps (step (iii)).

Hereinafter, the present invention is described further in detail by means of practicable examples.

EXAMPLE 1

In Example 1, the description is directed to a method including depositing GaN crystals on a sapphire substrate by a metal organic chemical vapor deposition (MOCVD) method, a molecular beam epitaxy (MBE) method, or a HVPE method and then obtaining a GaN-based single crystal substrate by a liquid phase epitaxy (LPE) method. The GaN-based crystal described herein denotes a semiconductor expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$). In the present example, the description is made with respect to the case of manufacturing a GaN single crystal substrate, as an example. However, for instance, an AlGaIn or AlN single crystal substrate also can be manufactured by a similar method.

First, as shown in FIG. 1A, a seed layer 12 of GaN is formed on a sapphire substrate 11 made of sapphire (crystalline Al_2O_3) by the MOCVD method. Specifically, the sapphire substrate is heated to have a temperature of about 1020° C. to 1100° C., and then trimethylgallium (TMG) and NH_3 are supplied onto the substrate to form the seed layer 12 of GaN. The Group III element contained in the seed layer 12 is not limited to gallium but may be aluminum or indium. That is, the seed layer 12 is not limited as long as it is formed of semiconductor crystals expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$).

Next, as shown in FIG. 1B, portions of the seed layer 12 located on its upper surface side are etched to form stripe-like convex portions. Specifically, first, a resist film is applied to the upper surface of the seed layer 12 and then the resist film thus applied is patterned into a stripe form by a photolithographic method, thus forming a resist pattern 13. Subsequently, with the resist pattern 13 used as a mask, the seed layer 12 is dry-etched and thereby in the upper part of the seed layer 12 is formed into a cyclic structure in which one cycle consists of a convex portion with a cross-sectional width of about 3 μm and a concave portion with a cross-sectional width of about 12 μm . In order to facilitate the separation of the substrate in a later step, it is preferable that the width of the convex portion is narrow. In the present example, the convex portion is formed to have a stripe structure. However, there is no problem even if it has another structure. For example, a dot structure may be provided as an in-plane structure.

Next, as shown in FIG. 1C, a mask film 14 is formed only in the concave portions. The mask film 14 may cover the whole side wall surfaces of the concave portions or only part of the wall surfaces. The mask film 14 is formed, for example, as follows. First, a thin film made of silicon nitride (SiN_x) is deposited to cover the seed layer 12 and the resist pattern 13 using an electron cyclotron resonance (ECR) sputter technique. In this case, solid silicon can be used as a raw material of silicon, nitrogen as a reactive gas, and argon as a plasma gas. In this manner, the formation of the mask film using the

ECR sputter technique allows a high-quality mask film to be formed at low temperature. Thereafter, the resist pattern 13 is subjected to lift-off and thereby the resist pattern 13 and the mask film 14 located on the resist pattern 13 are removed, which results in a state where the upper surfaces (C-planes) of the convex portions are exposed.

Next, as shown in FIG. 1D, with the upper surfaces of the convex portions thus exposed to serve as seed crystals, a selective growth layer 15 of GaN crystals is regrown. For example, GaN is formed by a low pressure MOCVD method (26600 Pa (200 Torr), 1050° C.). The selective growth layer 15 grows selectively from the upper surfaces of the convex portions that are not covered with the mask film 14. The selective growth layer 15 grows upward from the upper surface of each convex portion as well as in the direction parallel to the substrate surface (lateral growth). GaN that has grown in the lateral direction from the upper surface of a convex portion docks with GaN that has grown from a convex portion adjacent thereto, spatially (approximately in the middle of the concave portion concerned). Consequently, a bridging structure is formed. This allows the dislocation density to be reduced in the portions that have grown in the lateral direction.

With the selective growth layer 15 grown in such a manner, the respective crystals grown from the upper surfaces of a plurality of convex portions are combined together, and the upper surface thereof is a C-plane. As a result, the seed layer 12 and the selective growth layer 15 form a semiconductor layer including gaps. Consequently, a seed crystal substrate (a bridging structure substrate) 17 is obtained with the semiconductor layer including gaps formed therein. The Group III element contained in the selective growth layer 15 is not limited to gallium but may be aluminum or indium. That is, the selective growth layer 15 is not limited as long as it is formed of semiconductor crystals expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$).

Next, using the seed crystal substrate 17 obtained above, GaN crystals 16 are grown on the selective growth layer 15 by the LPE method (FIG. 1E). Thereafter, a part including the sapphire substrate 11 and a part including the GaN crystals 16 are separated from each other at the upper surfaces of the convex portions (FIG. 1F). In this manner, a substrate is obtained that is formed of GaN crystals alone.

The following description is directed to a method of growing the GaN crystals 16. FIG. 2A shows an example of an LPE apparatus used in this example.

The LPE apparatus shown in FIG. 2A includes: a raw material gas tank 21 for supplying nitrogen gas, or a mixed gas containing ammonia gas (NH_3 gas) and nitrogen gas, as a raw material gas; a pressure regulator 22 for regulating the pressure of a growth atmosphere; a leakage valve 23; a stainless steel container 24 used for crystal growth; and an electric furnace 25. FIG. 2B shows an enlarged view of the stainless steel container 24 inside of which a crucible 26 is set. The crucible 26 is made of boron nitride (BN), alumina (Al_2O_3), or the like. The crucible 26 can control its temperature within the range of 600° C. to 1000° C. The pressure (100 atm to 150 atm) of the atmosphere supplied from the raw material gas tank 21 can be controlled within the range of not higher than 100 atm by the pressure regulator 22.

A method of growing the GaN crystals is described below. First, normal amounts of Ga and Na serving as a flux were weighed and they were placed inside the crucible 26 together with the seed crystal substrate (i.e. the substrate shown in FIG. 1D). In the present example, the mole ratio of Ga and Na was set at 2.7:7.3. In order to make a comparison, a general

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seed crystal substrate having no bridging structure (i.e. a substrate with a GaN layer formed on a sapphire substrate) also was set at the same time to be subjected to the crystal growth. Subsequently, the crucible 26 was kept at 800° C. and nitrogen gas containing ammonia (40%) mixed therein was supplied at a pressure of 5 atm. Mixing of ammonia into the nitrogen gas can reduce the pressure of the atmosphere in which the crystal growth takes place but is not always required. Even in a nitrogen gas atmosphere free from ammonia, crystals can be grown under a pressure of 50 atm. In this state, while the temperature and pressure were kept constant, the LPE growth was carried out for 96 hours. The GaN crystals 16 thus obtained had a thickness of 100 μm .

The coefficient of linear expansion of sapphire used for the substrate is $7.5 \times 10^{-6}/\text{K}$ while that of the GaN that has grown is $5.5 \times 10^{-6}/\text{K}$. Hence, in a sample obtained through crystal growth carried out using a common seed crystal substrate, the substrate warped considerably at room temperature due to the high growth temperature (for instance, 800° C.). Such warping is a big problem in, for example, a step of mask alignment carried out in a device manufacturing process. On the other hand, in the sample with the GaN crystals grown on the seed crystal substrate (a bridging structure substrate) of the present invention, the substrate was separated at the interface between the seed layer 12 and the selective growth layer 15 when being cooled after the completion of crystal growth. Conceivably, this is caused by the stress generated by the difference in coefficient of linear expansion produced when the substrate is cooled after the completion of crystal growth and the concentration of the stress on the convex portions that is caused due to the small area of the convex portions. In this manner, a GaN single crystal substrate was obtained through the separation of the GaN crystals alone.

The GaN single crystals thus obtained were subjected to electron beam irradiation and thereby the in-plane distribution of cathodoluminescence (CL) was observed. Thus, the dislocation density thereof was evaluated. Dislocation parts are observed as black spots due to the low emission intensity thereof. The GaN single crystals obtained in the present example had a low dislocation density, specifically, $1 \times 10^2 \text{ cm}^{-2}$ or lower. On the other hand, the seed layer 12 disposed on the sapphire substrate 11 had a high dislocation density, specifically, $5 \times 10^9 \text{ cm}^{-2}$. Thus, growing the GaN single crystals by the method of the present example permitted the dislocation density of the crystals to be reduced.

The dislocation density of $1 \times 10^2 \text{ cm}^{-2}$ or lower is low and comparable to that of a GaAs substrate used in manufacturing semiconductor lasers for reading/writing information on/from CDs or DVDs. Hence, it is considered that according to the example described above, a GaN single crystal substrate was obtained that had sufficient characteristics for manufacturing semiconductor lasers having a lifetime of at least 10000 hours.

GaN crystals were produced by the method described above and then were subjected to measurements of dislocation density and PL intensity. They had a dislocation density of $1 \times 10^2 \text{ cm}^{-2}$ or lower. The spectrum of the PL intensity is shown in FIG. 3B. The intensity obtained at the peak around 360 nm of the spectrum shown in FIG. 3B was 22 V. FIG. 3A shows the PL intensity of a GaN thin film produced by a general MOCVD method for comparison. In this connection, FIGS. 3A and 3B show spectra measured with slits having different widths from each other, respectively. The peak intensity obtained around 360 nm of the spectrum shown in FIG. 3A was 0.48 V. The crystals obtained by the method of

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the present invention had a PL intensity that was about fifty times as high as that of the crystals produced by a conventional method.

In the present example, the description was directed to the manufacture of a GaN single crystal substrate using gallium. However, it is desirable to manufacture a substrate having low absorptivity with respect to a wavelength used in an optical device to be formed on the substrate. Hence, it is preferable that a substrate for a light emitting diode or a semiconductor laser handling light within the ultraviolet region is formed of $\text{Al}_x\text{Ga}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$) single crystals that contain a large amount of Al and have low absorptivity with respect to light with short wavelengths. In the present invention, it also is possible to substitute a part of Ga by another Group III element to form such Group III nitride semiconductor single crystals.

In the above-mentioned example, the GaN single crystal substrate was manufactured using a simple growth apparatus. However, it also is possible to manufacture substrates using a large growth apparatus in order to manufacture higher quality substrates. FIG. 4 shows an example of the LPE apparatus used in such a case.

The LPE apparatus shown in FIG. 4 includes an electric furnace 30 composed of a chamber 31 made of stainless steel and a furnace cover 32. The LPE apparatus is constructed to bear a pressure of 10 atm. Inside the chamber 31 are disposed heaters 33. The chamber 31 is composed of three zones that are zones 300a, 300b, and 300c provided with thermocouples 34a, 34b, and 34c, respectively. The three zones are controlled so as to keep their temperature within a range of $\pm 0.1^\circ \text{C}$., and thereby the temperature inside the furnace is controlled to be uniform. A core tube 35 is provided to improve the uniformity of the temperature inside the furnace and to prevent impurities from being introduced from the heaters 33.

Inside the core tube 35 is disposed a crucible 36 made of boron nitride (BN). A material is placed inside the crucible 36 and the temperature of the crucible 36 is raised to prepare a melt 37. A substrate 10 to serve as a seed crystal is attached to a substrate holding member 38. In the apparatus shown in FIG. 4, a plurality of substrates 10 can be attached to the substrate holding member 38. The substrate 10 is rotated by a rotary motor 39a. A stirring propeller 40 can be immersed in the melt 37. The propeller 40 is rotated by a rotary motor 39b. In the present example, a common rotary motor can be used since the atmosphere pressure is 10 atm or lower. However, an electromagnetic induction rotary mechanism may be used under an atmosphere pressure exceeding 10 atm. Atmosphere gas (raw material gas) is supplied from a gas source 41. The pressure of the atmosphere gas is adjusted by a pressure regulator 42. The atmosphere gas is fed into the furnace after impurities are removed therefrom by a gas purification unit 43.

The following description is directed to a method of growing crystals.

(1) First, predetermined amounts of Ga and Na to serve as a flux are weighed and then are placed inside a crucible. As Ga there is used a material with a purity of 99.9999% (six nine). As Na there is used purified Na. Na can be purified as follows: Na is heated to melt in a glove box, the inside of which has been subjected to He substitution, and then for example, oxides emerging on the surface layer are removed. Alternatively, Na may be purified by zone refining. The purity of Na can be increased by melting and solidifying Na repeatedly in a tube to allow impurities to deposit and then removing them.

(2) In order to melt the raw material placed inside the crucible, the temperature inside the electric furnace is raised to 900° C. and thereby a melt of the raw material is prepared.

In this stage, the seed crystal substrate is not placed inside the crucible. In order to mix Ga and Na, the melt is stirred for several hours with the propeller being immersed in the melt. The atmosphere gas is, for instance, nitrogen gas or that containing ammonia. The pressure of the nitrogen gas is set at about 1 atm to prevent Ga and Na from reacting with nitrogen gas at this stage. When ammonia is contained, the reaction occurs at lower pressure. Hence, it is preferable that the nitrogen gas alone is used for the atmosphere gas at this stage.

(3) Next, the temperature of the crucible is set at 800° C. to bring the melt into a supersaturation state. In addition, the atmosphere pressure is increased. In the present example, the atmosphere is, for example, nitrogen gas alone and is set at 50 atm. Subsequently, the seed crystal substrate is lowered to the position just above the surface of the melt and its temperature is made to approach the temperature of the melt. After several minutes, the seed crystal substrate is put into the melt to allow GaN crystals to start growing.

(4) During the crystal growth, the substrate is rotated at a rotational speed in a range of 10 rpm to 200 rpm. Desirably, the substrate is rotated at around 100 rpm. After the crystals are grown for 24 hours, the substrate is lifted to be taken out from the melt. After being lifted, the substrate is rotated at a speed of 300 rpm to 1500 rpm in order to remove the melt remaining on the substrate surface. Desirably, the substrate is rotated at around 1000 rpm. Thereafter, the substrate is taken out of the chamber. During the crystal growth, the temperature of the crucible may be kept constant. However, the temperature of the melt may be lowered at a constant rate to keep the supersaturation of the melt constant.

In the present example, since the seed crystal substrate having a bridging structure is used, the GaN substrate separates from the convex portions of the seed crystal substrate upon cooling the substrate. Consequently, a flat GaN single crystal substrate was obtained with no warping caused therein.

The substrate thus obtained was evaluated and as a result, its dislocation density was $1 \times 10^2 \text{ cm}^{-2}$ or lower. Furthermore, the PL intensity of the substrate thus obtained was about 50 times as high as that of a GaN thin film manufactured by a general MOCVD method.

In the present example was used a flux containing Na alone. However, similar effects can be obtained even when using a mixed flux containing alkaline-earth metal such as, for example, Ca and a Li, Na, or K flux. For instance, when using a mixed flux containing Na and Ca, the Ca mixed to account for about 10% of the whole allows crystals to grow at lower pressure.

According to the present invention, a GaN single crystal substrate that includes no sapphire substrate, is not warped, and has a low dislocation density can be manufactured with high mass-productivity. Accordingly, a substrate that permits a highly reliable device to be manufactured can be supplied at low cost. Particularly, its characteristics that no warping is caused and the dislocation density is low over the whole area simplify the process of manufacturing devices such as semiconductor lasers or the like and also allow devices to be manufactured with a high yield.

EXAMPLE 2

In Example 2, the description is directed to an example of manufacturing a semiconductor laser using the substrate obtained in Example 1. The configuration of a semiconductor laser 50 is shown in FIG. 5.

First, on a substrate 51 obtained in the above-mentioned example is formed a contact layer 52 of n-type GaN doped

with Si to have a carrier density of 5×10^{18} or lower. The substrate 51 is one with Group III nitride crystals formed on sapphire or one formed of Group III nitride crystals. In GaN-based crystals (crystals containing Ga and N), when Si is added thereto as an impurity, holes present in Ga increase in number. Since the holes present in Ga diffuse easily, they have harmful effects in terms of, for example, lifetime when a device is formed on the contact layer. Hence, the doping amount may be controlled so that the contact layer has a carrier density of 3×10^{18} or lower.

Next, on the contact layer 52 are formed a cladding layer 53 of n-type $\text{Al}_{0.07}\text{Ga}_{0.93}\text{N}$ and a light guiding layer 54 of n-type GaN. Subsequently, as an active layer 55 is formed a multiple quantum well (MQW) composed of a well layer (with a thickness of about 3 nm) made of $\text{Ga}_{0.8}\text{In}_{0.2}\text{N}$ and a barrier layer (with a thickness of about 6 nm) made of GaN. Thereafter, on the active layer 55 are formed sequentially a light guiding layer 56 of p-type GaN, a cladding layer 57 of p-type $\text{Al}_{0.07}\text{Ga}_{0.93}\text{N}$, and a contact layer 58 of p-type GaN. These layers can be formed by well-known methods. The semiconductor laser 50 is of a double-hetero junction type. The energy gap of the well layer containing indium in the MQW active layer is smaller than those of the n-type and p-type cladding layers containing aluminum. On the other hand, the highest optical refractive index is obtained in the well layer of the active layer 55, while the light guiding layers and the cladding layers have optical refractive indices decreased in this order.

On the contact layer 58 is formed an insulating film 59 forming a current injection region having a width of about 2 μm . A ridge part to serve as a current constriction part is formed in the upper portion of the p-type cladding layer 57 and the p-type contact layer 58.

On the upper side of the p-type contact layer 58 is formed a p-side electrode 500 that is in ohmic contact with the contact layer 58. The p-side electrode 500 may be formed of a layered product of nickel (Ni) and gold (Au).

On the n-type contact layer 52 is formed an n-side electrode 501 that is in ohmic contact with the contact layer 52. The n-side electrode 501 may be formed of a layered product of titanium (Ti) and aluminum (Al).

The semiconductor laser produced by the method described above was subjected to a device evaluation. When predetermined forward voltage was applied to between the p-side electrode and the n-side electrode of the semiconductor laser obtained in the above, positive holes and electrons are injected into the MQW active layer from the p-side electrode and the n-side electrode, respectively, to be recombined with each other in the MQW active layer and thereby an optical gain was produced. Consequently, the semiconductor laser generated oscillation with an emission wavelength of 404 nm.

In the semiconductor laser of the present example, a substrate with a low dislocation density, specifically, $1 \times 10^2 \text{ cm}^{-2}$ or lower, was used as its substrate. Accordingly, the semiconductor laser had a lower threshold, improved luminous efficiency, and improved reliability as compared to one produced on a GaN substrate with a high dislocation density.

Effects obtained when LDs or LEDs are manufactured using the substrate obtained according to the present invention are described as follows. The sapphire substrate can be removed and thereby a GaN single crystal substrate with no warping and a low dislocation density is used. This allows, for example, mask alignment to be carried out easily and accurately in a LD manufacturing process. Consequently, highly reliable LDs can be manufactured with a high yield. Furthermore, since the use of the GaN single crystal substrate with no warping and a low dislocation density reduces the effect

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causing distortion of a device, the reliability of the device can be improved. Since the dislocation density is low over the entire area of the substrate, high reliability can be obtained in wide-stripe LDs. Moreover, the use of the GaN substrate makes cleavage possible and thereby the dry etching step is no longer necessary. Consequently, the manufacturing process can be simplified and thus devices can be manufactured at low cost.

EXAMPLE 3

In Example 3, the description is directed to an example of a method of forming concavities and convexities in a substrate. First, as shown in FIG. 6A, concave and convex portions to form gaps are formed on a sapphire substrate 61 made of sapphire (crystalline Al_2O_3) having a (0001) plane. A stripe resist pattern is formed by photolithography and then the concave and convex portions are formed by dry-etching using chlorine gas.

Next, as shown in FIG. 6B, a SiNx film 62 is formed only in the concave portions to be gaps. Specifically, first, a resist pattern is formed on the convex portions, and then a thin film made of silicon nitride (SiNx) is deposited using an electron cyclotron resonance (ECR) sputter technique. By lift-off of the resist pattern, a SiNx film is formed in the concave portions.

Subsequently, a seed layer 63 of GaN is formed on the convex portions of the sapphire surface by the MOCVD method. Specifically, after the sapphire substrate is heated to have a temperature of about 1020°C . to 1100°C ., trimethylgallium (TMG) and NH_3 are supplied onto the substrate to form the seed layer of GaN. In this connection, the Group III element of the seed layer is not limited to gallium but may be aluminum or indium. In other words, any seed layer can be used as long as it is formed of semiconductor crystals expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$). With the crystal layer of the Group III nitride semiconductor thus obtained used as a seed crystal, crystal growth is carried out.

Example 4

In Example 1, the GaN single crystals were formed on a seed crystal substrate having a bridging structure by the LPE method using a flux. However, the substrate having a bridging structure also can be utilized as a seed crystal used for growing crystals by a sublimation method. An example of the crystal growing apparatus used in this case is shown in FIG. 7.

The crystal growing apparatus shown in FIG. 7 includes a flow-controller 80, an electric furnace 81, and a thermocouple 87. Inside the electric furnace 81 is disposed a silica tube 82 in which a crucible 84 made of boron nitride (BN) is placed with GaN powder 83 contained therein. Nitrogen gas containing NH_3 gas is supplied from the direction indicated with an arrow 88 in FIG. 7 and the atmosphere inside the electric furnace 81 is adjusted by the flow-controller 80 so as to be a nitrogen gas atmosphere containing about 10 vol. % of NH_3 gas. When the temperature of the crucible 84 is raised, the GaN powder 83 reacts with the nitrogen gas (NH_3 gas) to be decomposed and springs out upward to adhere onto the substrate 86 that has been heated by a substrate heater 85. For a substrate to be placed on the substrate heater 85 there is used the seed crystal substrate having a bridging structure described in Example 1. After GaN single crystals were grown on the GaN seed crystal substrate, the temperature inside the electric furnace was lowered and the temperature of the substrate heater also was lowered. As a result, the GaN

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single crystals that had grown were separated from the sapphire substrate due to the difference in coefficient of linear expansion between the sapphire substrate and the GaN crystals.

The pressure of the electric furnace is preferably 1 atm or higher. As an increase of the pressure in the electric furnace causes a rise of the decomposition temperature of GaN and facilitates the decomposition of GaN in the crucible and production of GaN on the substrate, GaN crystals can be grown stably on the substrate.

In the method of the example described above, a C-plane $\text{Al}_x\text{Ga}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$) substrate can be used as a seed crystal. However, the use of an $\text{Al}_x\text{Ga}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$) substrate having another orientation, as a seed crystal substrate also makes it possible to obtain a single crystal substrate expressed by a composition formula of $\text{Al}_x\text{Ga}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$). For example, when with an A-plane GaN substrate used as a seed crystal, a light-emitting diode is formed using the single crystal substrate obtained, no piezoelectric effect is produced. Consequently, positive holes and electrons can be recombined efficiently and thereby the luminous efficiency can be improved.

Using a substrate obtained by a manufacturing method of the present invention, Group III nitride crystals are grown epitaxially on the substrate, and thereby a semiconductor device is obtained that includes a semiconductor element such as, for example, an LD or LED.

EXAMPLE 5

First, as shown in FIG. 8A, a seed layer 92 of GaN is formed on a sapphire substrate 91 made of sapphire (crystalline Al_2O_3) by the MOCVD method. Specifically, the sapphire substrate 91 is heated to have a temperature of about 1020°C . to 1100°C . Thereafter, trimethylgallium (TMG) and NH_3 are supplied onto the substrate to form the seed layer 92 of GaN. The Group III element of the seed layer 92 is not limited to gallium but may be aluminum or indium. That is, the seed layer 92 may be any layer as long as it is formed of semiconductor crystals expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$ and $0 \leq v \leq 1$).

Next, as shown in FIG. 8B, the seed layer 92 is etched partially up to the sapphire substrate 91 to form stripe convex portions. Specifically, first, after a resist film 93 is applied to the upper surface of the seed layer 92, the resist film 93 thus applied is patterned into stripes by a photolithographic method to form a resist pattern. Subsequently, with the resist pattern used as a mask, the seed layer 92 is dry-etched. Consequently, as shown in FIG. 8C, convex portions with a width of about $5\text{ }\mu\text{m}$ are formed at a cycle of about $300\text{ }\mu\text{m}$. The convex portions each are formed of a stripe structure in the present example. However, there is no problem even if the convex portions have other structures. For instance, dot-like structures may be provided as in-plane structures.

Next, as shown in FIG. 8D, LPE-GaN crystals 94 each formed of a GaN crystal are grown through liquid phase growth, with the upper surfaces of the convex portions used as seed crystals. The liquid phase growth was carried out using the LPE apparatus shown in FIGS. 2A and 2B. Sodium and gallium were weighed and then were placed inside the crucible into which a template shown in FIG. 8C was inserted. Crystals were grown in a nitrogen pressurized atmosphere with a pressure of 50 atm at 800°C . for 100 hours. Consequently, the LPE-GaN crystals shown in FIG. 8D were grown. In the liquid phase growth, since the growth rate in the lateral

direction is faster, the LPE-GaN crystals that grew from the convex portions adjoining each other coalesced as shown in FIG. 8D.

In the present example, the area where the convex portions and the LPE-GaN crystals are in contact with each other is small. Hence, when a melt was cooled, the LPE-GaN crystals were separated easily in the vicinities of the gaps. Furthermore, in the LPE-GaN crystals thus obtained, a great deal of dislocations were observed in their portions that had been located on the convex portions and coalescence portions, but in the other portions fewer dislocations were observed. In the present invention, since the cycle of the convex portions is 300 μm , a region with less dislocations can be obtained in a wide area, specifically, at least 100 μm . Accordingly, when, for example, a semiconductor laser is manufactured, the precision required in mask alignment carried out for forming a waveguide can be eased. In addition, a waveguide of a wide stripe required for a high-power semiconductor laser can be formed. Hence, great practical effects are provided.

Furthermore, as another example, even when using a template shown in FIG. 9A, similar LPE-GaN crystals can be formed.

As shown in FIG. 9A, first, SiNx (silicon nitride) to form a mask film 102 is grown on a sapphire substrate 101 to have a thickness of 100 nm by an atmospheric CVD method. Subsequently, stripe windows (portions where the sapphire substrate is exposed) are formed in the mask film by photolithography and etching. The windows may be in the form of dot. The mask film thus left had a width of 10 μm and was formed at a cycle of 500 μm .

Next, as shown in FIG. 9B, on the portions of the sapphire substrate that are not covered with the mask film 102 a seed layer 103 expressed by a composition formula of $\text{Al}_u\text{Ga}_v\text{In}_{1-u-v}\text{N}$ (wherein $0 \leq u \leq 1$, $0 \leq v \leq 1$, and $u+v \leq 1$) is formed by the MOCVD method. In this example, after the sapphire substrate is heated to have a temperature of about 1020° C. to 1100° C., trimethylgallium (TMG) and NH_3 are supplied onto the substrate to grow the seed layer.

Next, as shown in FIG. 9C, LPE-GaN crystals 104 each formed of a GaN crystal are grown on the seed crystals through liquid phase growth. The liquid phase growth was carried out using the LPE apparatus shown in FIGS. 2A and 2B.

Similarly, in the present example, the area where the convex portions and the LPE-GaN crystals were in contact with each other was small. Hence, when the melt was cooled, the LPE-GaN crystals were separated easily in the vicinities of the gaps. Furthermore, in the LPE-GaN crystals thus obtained, a great deal of dislocations were observed in their portions that had been located on the convex portions and coalescence portions, but in the other portions fewer dislocations were observed. Hence, when, for example, a semiconductor laser is manufactured, the precision required in mask alignment carried out for forming a waveguide can be eased. In addition, a waveguide of a wide stripe required for a high-power semiconductor laser can be formed. Consequently, great practical effects are provided.

EXAMPLE 6

As shown in FIGS. 10A to 10C, crystal growth is carried out using a metalorganic vapor phase epitaxy (MOVPE) method. That is, prior to vapor phase growth, a sapphire C-plane substrate 111 is placed on a susceptor located inside a reactor and is heated in a hydrogen atmosphere of 200 Torr at 1050° C. for 15 minutes after the reactor is evacuated of air, and thereby the substrate surface is cleaned.

Subsequently, after the substrate is cooled to 600° C., trimethylgallium (TMG) is allowed to flow at 20 $\mu\text{mol/min}$, ammonia at 2.5 L/min, carrier hydrogen and nitrogen each at 2 L/min. As a result, a GaN buffer layer 112 is deposited in a polycrystalline state to have a thickness of 20 nm as shown in FIG. 10A. In this case, the optimum deposition temperature is 500° C. to 600° C.

Next, only the supply of TMG is stopped and the substrate temperature is raised to 1090° C. Thereafter, TMG is supplied again to form a GaN single crystal layer 113 with a thickness of about 1 μm . It is desirable that the thickness is at least 0.5 μm in order to improve the c-axis orientation. A preferable growth temperature range is higher than 1000° C. but not higher than 1200° C.

Next, the supply of TMG and hydrogen is stopped, and then the substrate temperature is lowered to 800° C. in an atmosphere of a mixture of ammonia and nitrogen to allow the substrate to have a constant temperature. Thereafter, trimethylindium (TMI) and TMG are supplied at 200 $\mu\text{mol/min}$ and 20 $\mu\text{mol/min}$, respectively, to deposit an InGaN layer 114 having a thickness of 100 nm. The mole fraction of In contained in the InGaN mixed crystals is about 10%. It is possible to adjust the mole fraction of In contained in the InGaN mixed crystals by adjusting the mole fractions of TMI and TMG to be supplied.

Subsequently, the supply of TMI and TMG is stopped and the substrate temperature is raised from 800° C. to 1090° C. in the atmosphere of a mixture of ammonia and nitrogen. The temperature is raised within about 3 to 5 minutes, i.e. within a relatively short time. In this case, as shown in FIG. 10B, concavities and convexities having a diameter and a depth on the order of several tens of nanometers are produced at the entire surface of the InGaN layer 114. Conceivably, this is because InN crystals having a vapor pressure that is orders of magnitude greater than that of the InGaN layer evaporate from the InGaN active layer during the temperature increase.

Next, TMG and carrier hydrogen are added at 20 $\mu\text{mol/min}$ and 2 L/min, respectively, and thereby a GaN single crystal layer 115 is grown to have a thickness of about 1 μm as shown in FIG. 10C. Thereafter, only the supply of TMG is stopped, and the substrate is cooled to room temperature in an atmosphere containing ammonia, hydrogen, and nitrogen.

As a result of the growth carried out by the steps described above, gaps having a diameter and a depth on the order of several tens of nanometers can be produced at the interface between the InGaN single crystal layer 114 and the GaN single crystal layer 115 located at the top as shown in FIG. 10C.

In the present example, in order to produce gaps at the interface between the GaN single crystal layer 115 and the InGaN single crystal layer 114 located on the substrate side thereof, the InGaN layer 114 was heat-treated. However, it is to be understood that in the case of an AlGaInN layer containing In, the same effect can be obtained. As the method of forming concavities and convexities at the surface of the InGaN layer 114 to produce gaps, a heat treatment is useful, and particularly, rapid temperature increase is effective. A mole fraction of In contained in the InGaN layer 114 is preferably at least 10% but is not limited as long as In is contained. Furthermore, preferably, the film thickness is at least 100 nm. However, gaps can be obtained when the thickness is at least 10 nm.

In the present example, the description was made with respect to the case of vacuum growth. However, similar effects can be obtained by the growth carried out in an atmospheric or pressurized atmosphere.

Using the seed substrate described above, Group III nitride crystals are grown in a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen. When this seed substrate is used, the crystals are separated easily in the vicinities of the gaps, so that a substrate is obtained easily that is formed of Group III nitride crystals alone.

EXAMPLE 7

As shown in FIGS. 11A to 11C, crystal growth is carried out using the metal organic chemical vapor deposition (MOCVD) method. In other words, prior to the vapor deposition, a sapphire C-plane substrate 111 is placed on a susceptor located inside a reactor and is heated in a hydrogen atmosphere of 200 Torr at 1050° C. for 15 minutes after the reactor is evacuated of air, and thereby the substrate surface is cleaned.

Subsequently, after the substrate is cooled to 600° C., trimethylgallium (TMG) is allowed to flow at 20 μmol/min, trimethylindium (TMI) at 200 μmol/min, ammonia at 2.5 L/min, and carrier nitrogen at 2 L/min. As a result, an InGaN buffer layer 116 is deposited in a polycrystalline state to have a thickness of 20 nm to 100 nm as shown in FIG. 11A. In this case, the optimum deposition temperature is 500° C. to 600° C.

Subsequently, the supply of TMG and TMI is stopped and the substrate temperature is raised to 1090° C. The temperature is raised within about 3 to 5 minutes, i.e. within a relatively short time. In this case, as shown in FIG. 11B, concavities and convexities having a diameter and a depth on the order of several tens of nanometers are produced at the entire surface of the InGaN buffer layer 116. Next, carrier hydrogen is added to flow at 2 L/min and then TMG is supplied again, and thereby a GaN single crystal layer 117 is formed to have a thickness of about 1 μm as shown in FIG. 11C. It is desirable that the thickness is at least 0.5 μm in order to improve the c-axis orientation. A preferable growth temperature range is higher than 1000° C. but not higher than 1200° C.

Finally, only the supply of TMG is stopped and the substrate is cooled to room temperature in an atmosphere containing ammonia, hydrogen, and nitrogen.

As a result of the growth carried out by the steps described above, gaps having a diameter and a depth on the order of several tens of nanometers can be produced at the interface between the InGaN buffer layer 116 and the GaN single crystal layer 117 located at the top as shown in FIG. 11C.

In the present example, in order to produce gaps at the interface between the GaN single crystal layer 117 and the InGaN buffer layer 116 located on the substrate side thereof, the InGaN buffer layer 116 was heat-treated. However, it is to be understood that in the case of an AlGaInN buffer layer containing In, similar effects can be obtained. As the method of forming concavities and convexities at the surface of the InGaN buffer layer 116 to produce gaps, a heat treatment is useful, and particularly, rapid temperature increase is effective. A mole fraction of In contained in the InGaN buffer layer 116 is preferably at least 10% but is not limited as long as In is contained. Furthermore, preferably, the film thickness is at least 100 nm. However, gaps can be obtained when the thickness is at least 10 nm.

Using the seed substrate described above, Group III nitride crystals are grown in a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen. When this seed substrate is used, the crystals are separated easily in the

vicinities of the gaps, so that a substrate is obtained easily that is formed of Group III nitride crystals alone.

According to the manufacturing methods of the present invention, substrates can be manufactured easily that has less warping and is formed of high-quality Group III nitride crystals alone.

The invention may be embodied in other forms without departing from the spirit or essential characteristics thereof. The embodiments disclosed in this application are to be considered in all respects as illustrative and not limiting. The scope of the invention is indicated by the appended claims rather than by the foregoing description, and all changes which come within the meaning and range of equivalency of the claims are intended to be embraced therein.

What is claimed is:

1. A method of manufacturing a Group III nitride substrate, the method comprising:

- (i) forming a Group III nitride layer having a cycle of gaps of at least 100 μm, on a substrate;
- (ii) bringing a surface of the Group III nitride layer into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen, to make the at least one Group III element and the nitrogen react with each other to grow Group III nitride crystals on the Group III nitride layer; and
- (iii) separating a part including the substrate and a part including the Group III nitride crystals from each other in vicinities of the gaps.

2. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the at least one Group III element is gallium, and the Group III nitride crystals are GaN crystals.

3. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the atmosphere containing nitrogen is a pressurized atmosphere.

4. The method of manufacturing a Group III nitride substrate according to claim 1, wherein in the process (iii), separation is carried out using stress generated by a difference in coefficient of linear expansion between the substrate and the Group III nitride crystals.

5. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the process (i) comprises:

- (i-1) forming a first semiconductor layer expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$), on the substrate;
- (i-2) forming convex portions by partially removing the first semiconductor layer; and
- (i-3) forming the Group III nitride layer having gaps in its portions other than the convex portions by growing a second semiconductor layer from upper surfaces of the convex portions of the first semiconductor layer, the second semiconductor layer being expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$),

wherein in the process (iii), the first semiconductor layer and the second semiconductor layer are separated from each other at the upper surfaces of the convex portions.

6. The method of manufacturing a Group III nitride substrate according to claim 5, wherein the upper surfaces are C-planes.

7. The method of manufacturing a Group III nitride substrate according to claim 5, wherein in the process (i-2), the convex portions are formed in stripes.

8. The method of manufacturing a Group III nitride substrate according to claim 5, wherein in the process (i-2),

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concave portions that are portions other than the upper surfaces of the convex portions are covered with a mask film.

9. The method of manufacturing a Group III nitride substrate according to claim 8, wherein the mask film contains at least one selected from a group consisting of silicon nitride, oxide silicon, nitride oxide silicon, aluminum oxide, aluminum nitride oxide, titanium oxide, zirconium oxide, and niobium oxide.

10. The method of manufacturing a Group III nitride substrate according to claim 8, wherein the mask film is made of high melting metal or a high melting metallized material.

11. The method of manufacturing a Group III nitride substrate according to claim 8, wherein the mask film contains at least one selected from a group consisting of tungsten, molybdenum, niobium, tungsten silicide, molybdenum silicide, and niobium silicide.

12. A method of manufacturing a Group III nitride substrate, comprising:

(I) forming convex portions by processing a surface of a substrate;

(II) growing a Group III nitride layer from upper surfaces of the convex portions to form a seed crystal substrate having a cycle of gaps of at least 100 μm , formed between the substrate and the Group III nitride layer;

(III) bringing a surface of the Group III nitride layer into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in a pressurized atmosphere containing nitrogen, to make the at least one Group III element and the nitrogen react with each other to grow Group III nitride crystals on the Group III nitride layer; and

(IV) separating a part including the substrate and a part including the Group III nitride crystals from each other in vicinities of the gaps.

13. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the substrate is a sapphire substrate.

14. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the alkali metal is at least one selected from sodium, lithium, and potassium.

15. The method of manufacturing a Group III nitride substrate according to claim 1, wherein the melt further contains alkaline-earth metal.

16. A method of manufacturing a Group III nitride substrate, the method comprising:

(i) forming a Group III nitride layer having a cycle of gaps of at least 100 μm , on a substrate;

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(ii) bringing a surface of the Group III nitride layer into contact with a melt containing alkali metal and at least one Group III element selected from gallium, aluminum, and indium, in an atmosphere containing nitrogen, to make the at least one Group III element and the nitrogen react with each other to grow Group III nitride crystals on the Group III nitride layer; and

(iii) separating a part including the substrate and a part including the Group III nitride crystals from each other in vicinities of the gaps;

wherein in the process (i), the Group III nitride layer including gaps includes a semiconductor layer expressed by a composition formula of $\text{Al}_x\text{Ga}_y\text{In}_{1-x-y}\text{N}$ (wherein $0 \leq x \leq 1$ and $0 \leq y \leq 1$), and after forming the semiconductor layer, the gaps are formed in the semiconductor layer or at a surface of the semiconductor layer through a temperature-programmed heat treatment carried out in an atmosphere of a mixture of ammonia and nitrogen.

17. The method of manufacturing a Group III nitride substrate according to claim 16, wherein the Group III nitride layer including gaps is a semiconductor layer expressed by a composition formula of $\text{Ga}_x\text{In}_{1-x}\text{N}$ (wherein $0 \leq x \leq 1$).

18. The method of manufacturing a Group III nitride substrate according to claim 16, wherein the temperature-programmed heat treatment is carried out at a programming rate of 50 to 100° C./min.

19. The method of manufacturing a Group III nitride substrate according to claim 16, wherein a cycle of the gaps is at least 30 μm .

20. The method of manufacturing a Group III nitride substrate according to claim 16, wherein a cycle of the gaps is at least 50 μm .

21. The method of manufacturing a Group III nitride substrate according to claim 16, wherein a cycle of the gaps is at least 100 μm .

22. A Group III nitride substrate manufactured by a manufacturing method according to claim 1.

23. The Group III nitride substrate according to claim 22, wherein a cycle of dense dislocation areas is at least 30 μm .

24. The Group III nitride substrate according to claim 22, wherein a cycle of dense dislocation areas is at least 50 μm .

25. The Group III nitride substrate according to claim 22, wherein a cycle of dense dislocation areas is at least 100 μm .

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